



Development of High-performance Niobium-3 Tin Cavities at Cornell University

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June 28, 2023

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Grant No. PHY-1549132

Grant No. DE-SC0008431

This work made use of the Cornell Center for Materials Research Shared Facilities which are supported through the NSF MRSEC program (DMR1719875).

This work was performed in part at the Cornell NanoScale Facility, a member of the National Nanotechnology Coordinated Infrastructure (NNCI), which is supported by the National Science Foundation (Grant NNCI-2025233)

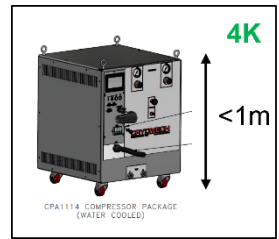
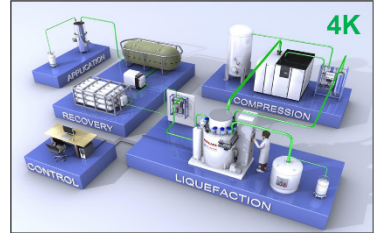
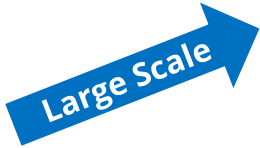
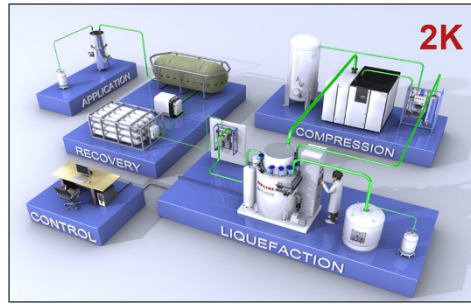
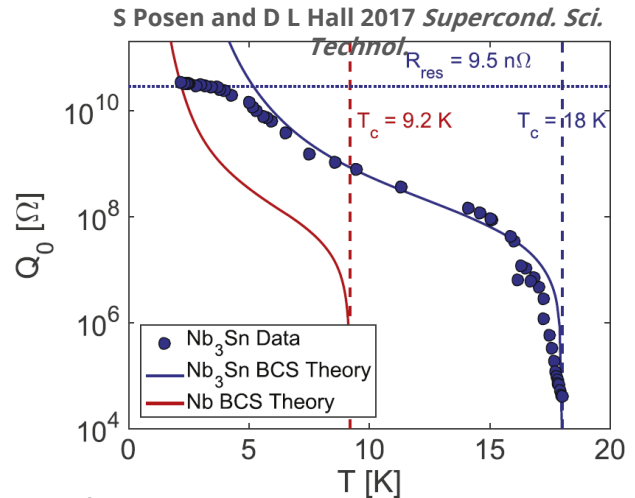
Nb₃Sn Potential

	Nb	Nb ₃ Sn
Critical Temperature (T_c)	9 K	18 K
Q_0 at 4.2K	6×10^8	6×10^{10}
Superheating Field (B_{sh})	240 mT	420 mT
Max. E_{acc}	55 MV/m	100 MV/m

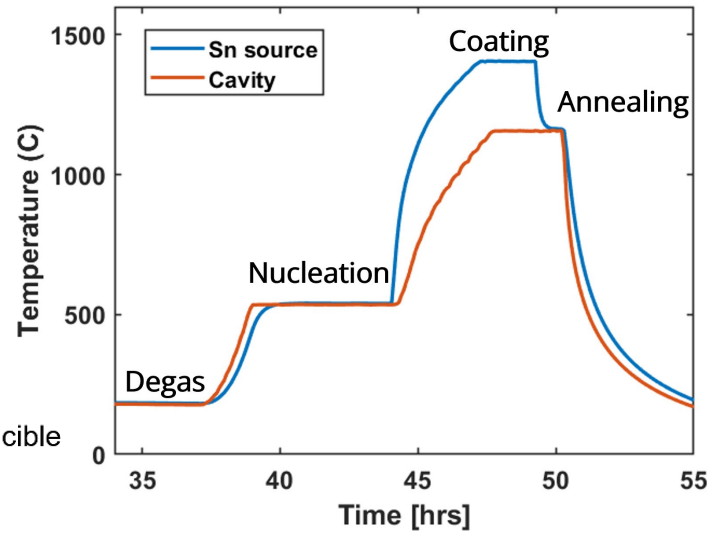
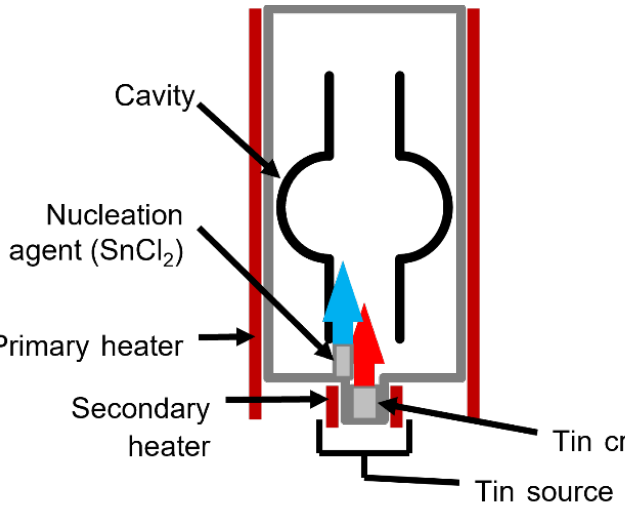
Higher operating temperature
 ≡ Lower cooling cost and complexity

Higher superheating field B_{sh} = higher
 accelerating gradients: $E_{acc,max} \propto B_{sh}$

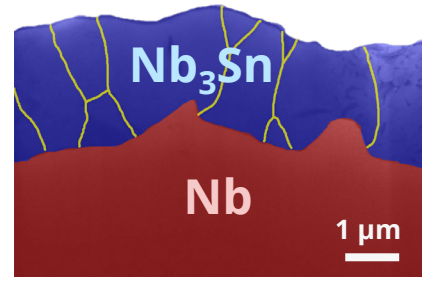
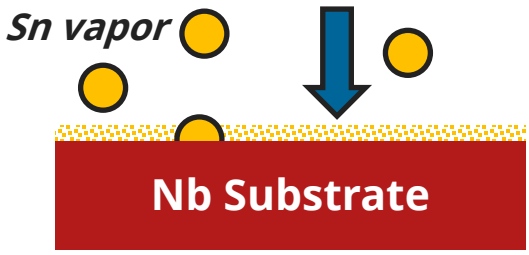
Q_0 given for 1.3 GHz single-cell ILC-shape cavities



State-of-the-Art: Thermal Vapor Diffusion



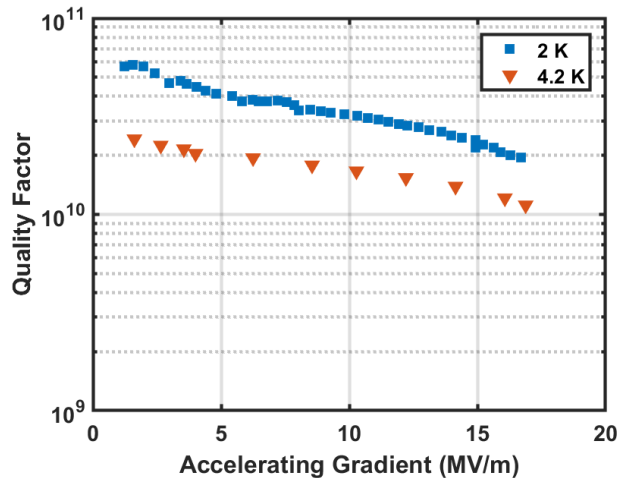
“Wuppertal” configuration
(secondary heater for the tin source)



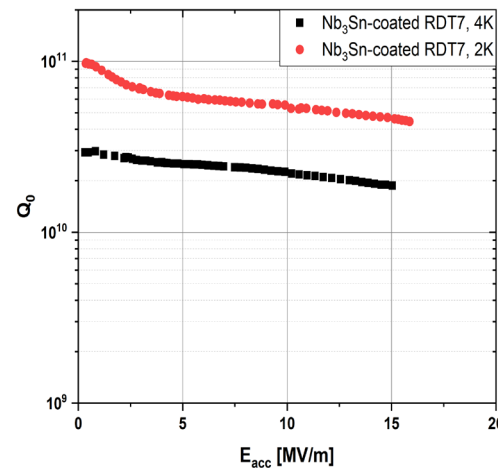
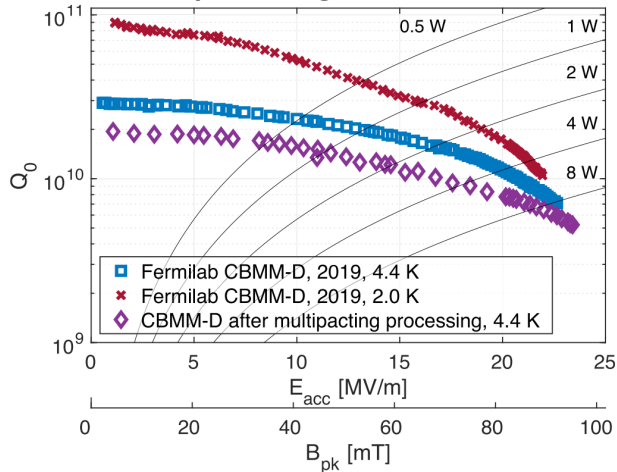
S. Posen and M. Liepe, Phys. Rev. ST Accel. Beams 15, 112001 (2014)

State-of-the-art Performance

- Reproducible performance of ~ 4 K operation with $Q_0 > 10^{10}$ at typical CW operating fields.
- Current quench fields: 16-24 MV/m (FNAL holds world record)



S. Posen, <https://doi.org/10.1088/1361-6668/abc7f7>



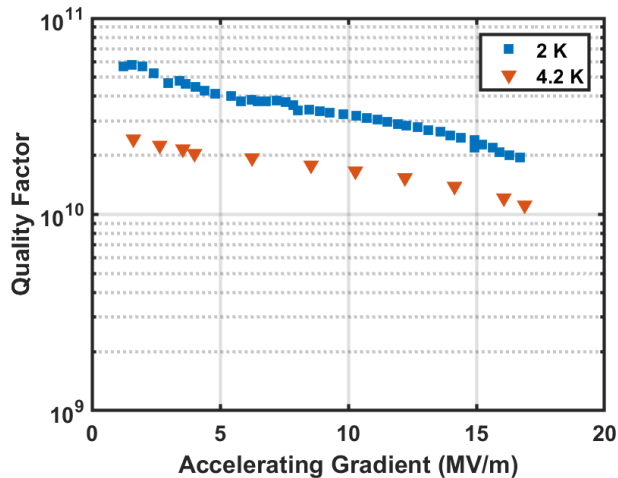
(Note: Plots are for single cell cavities)

State-of-the-art Performance

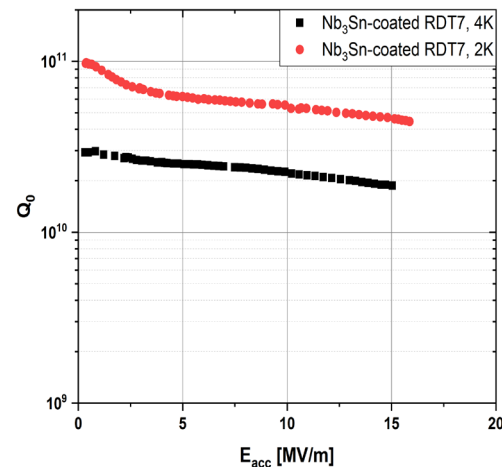
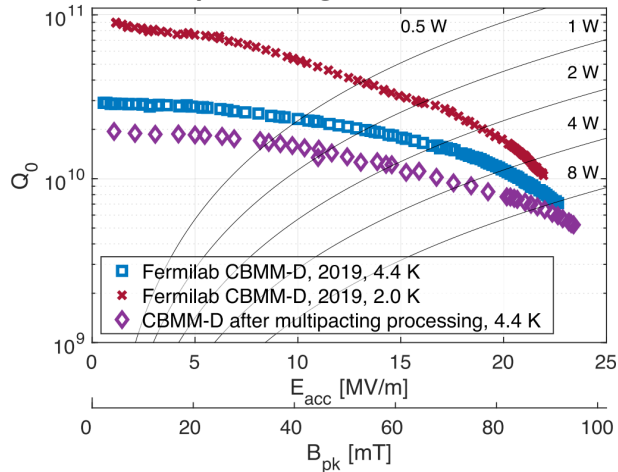
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Cornell University



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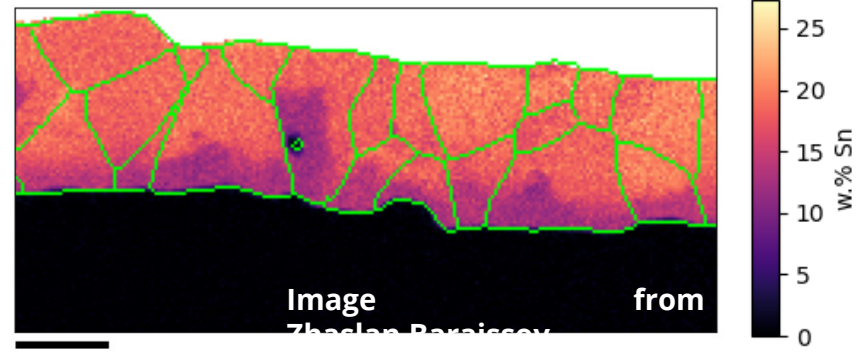


Efforts continue worldwide to push the limits of Nb₃Sn even further!

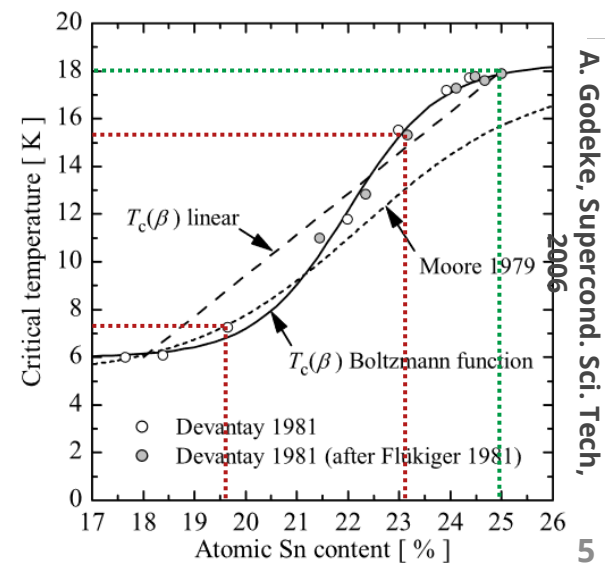
Some presentations here at SRF'23: TUPTB009, TUPTB010, TUPTB014, TUPTB015, TUPTB020, TUPTB065, WEPWB126

1. Tin depleted regions

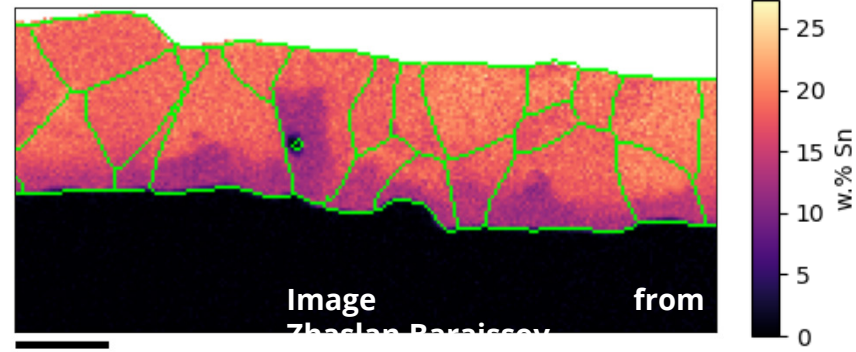
→ Tin depleted regions have a **low critical temperature**



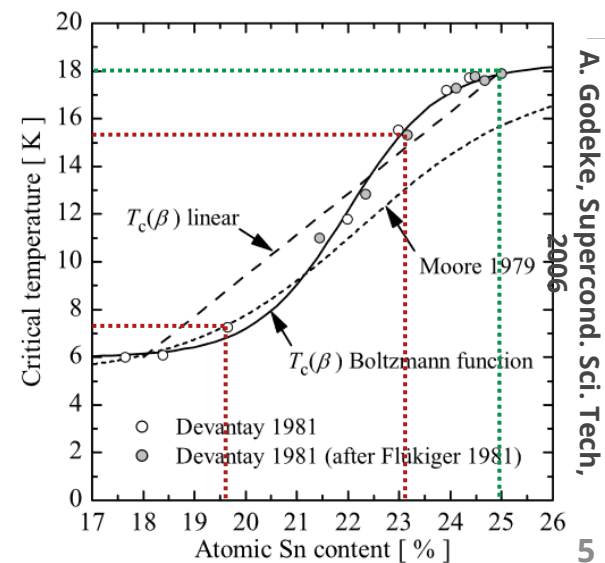
1 μm



- 1. Tin depleted regions
 - Tin depleted regions have a **low critical temperature**
- 2. Uniformity of thickness of film
 - Thinner patches expose Nb-Nb₃Sn interface (tin depleted)



1 μm



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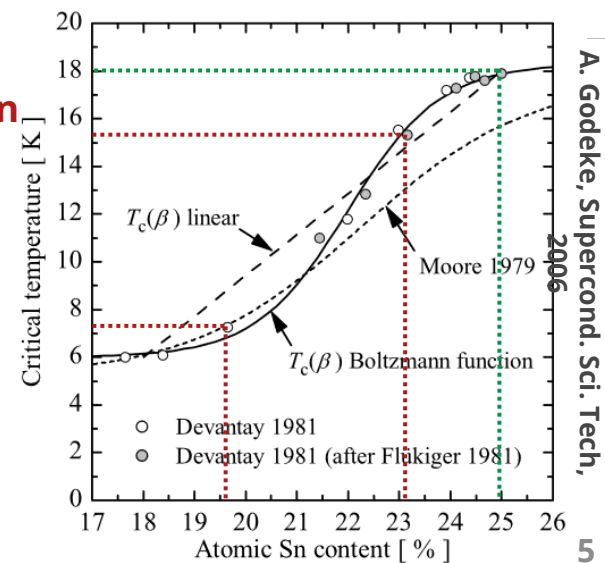
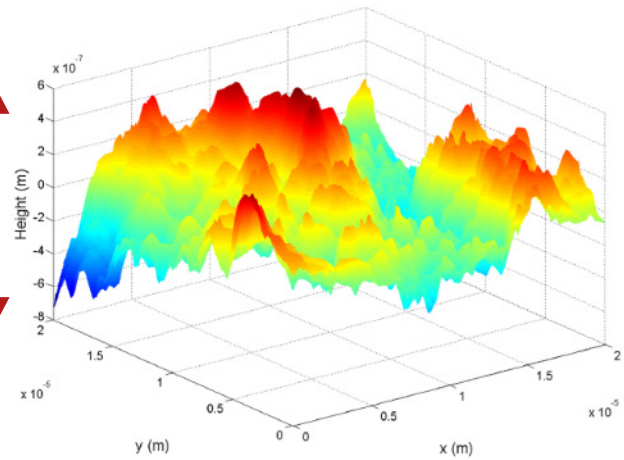
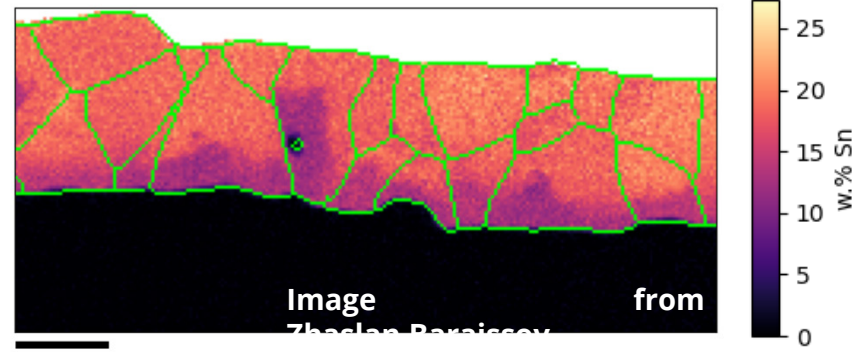
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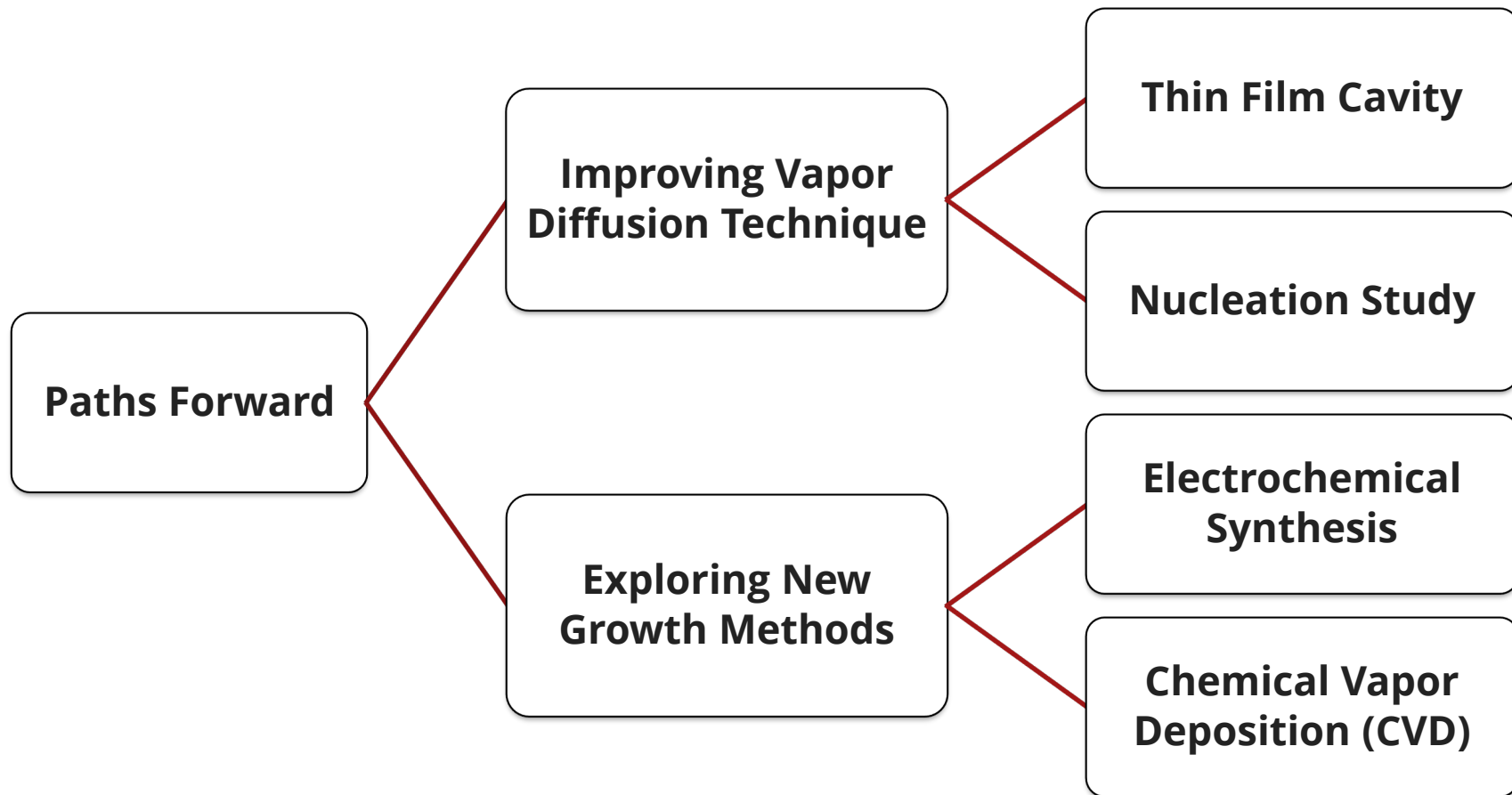
3. Surface roughness

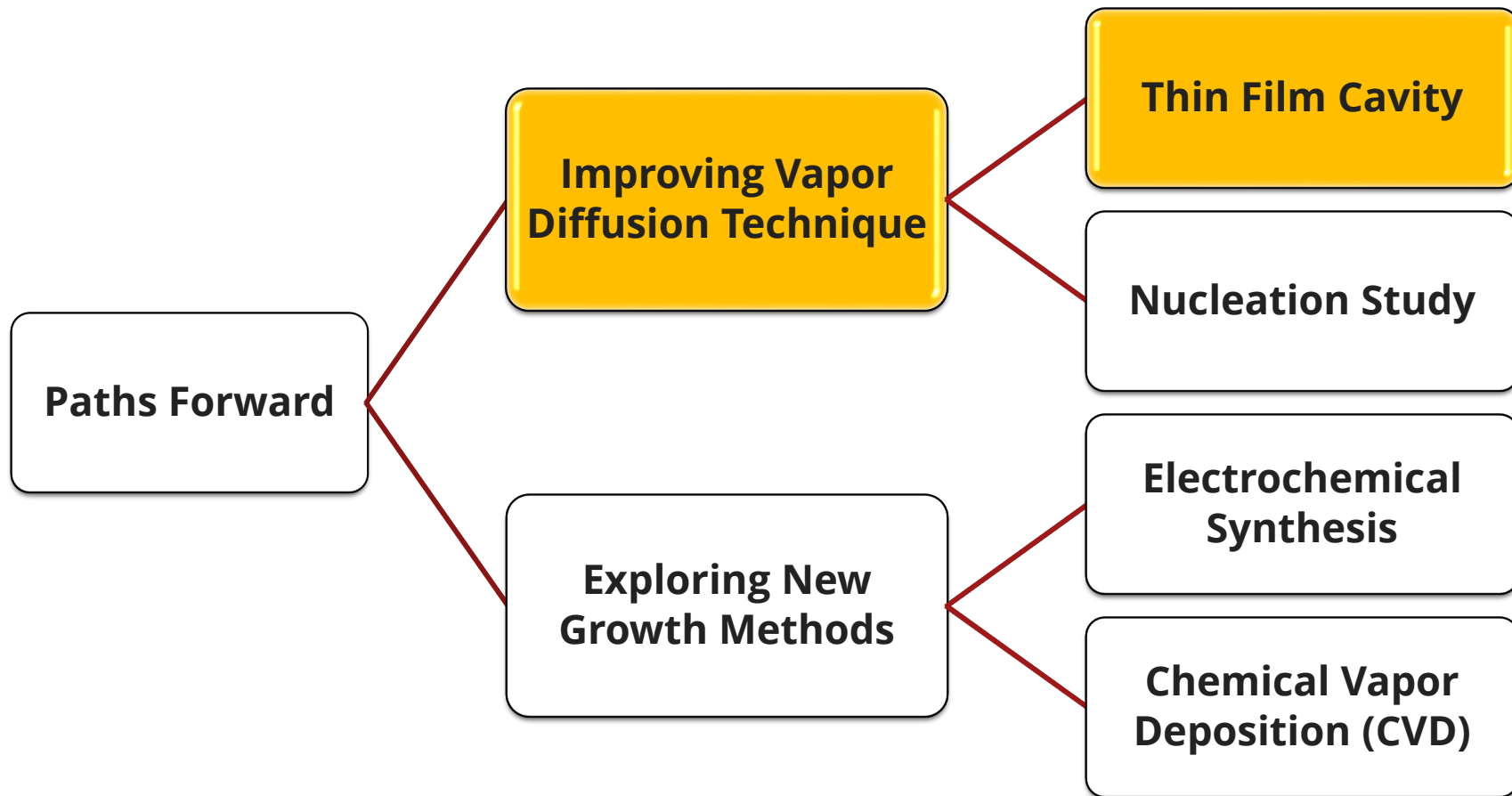
→ Can cause **field enhancement and vortex nucleation**



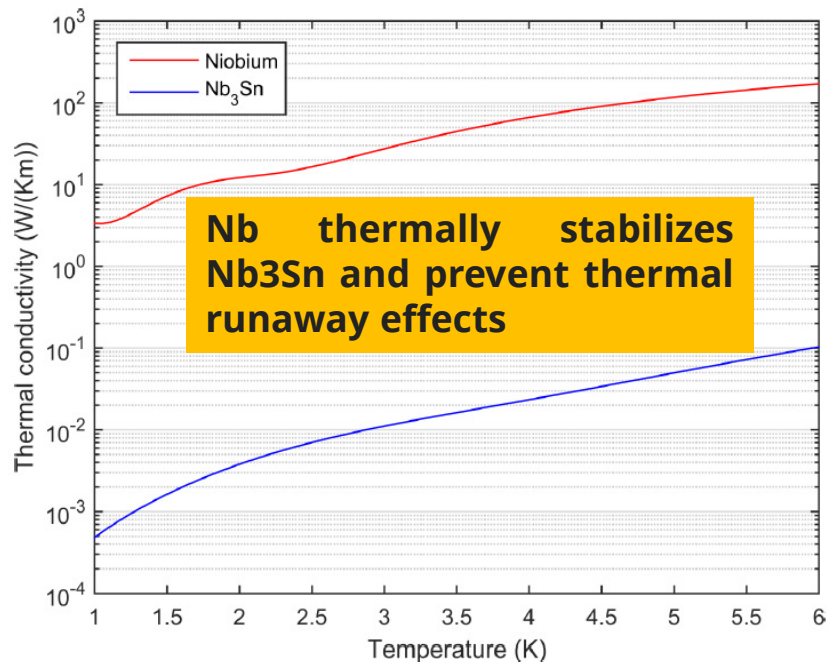
R. D. Porter, Ph.D. thesis,
Phys. Dept., Cornell
University, Ithaca, United
States, 2021

A. Godeke, Supercond. Sci. Tech,

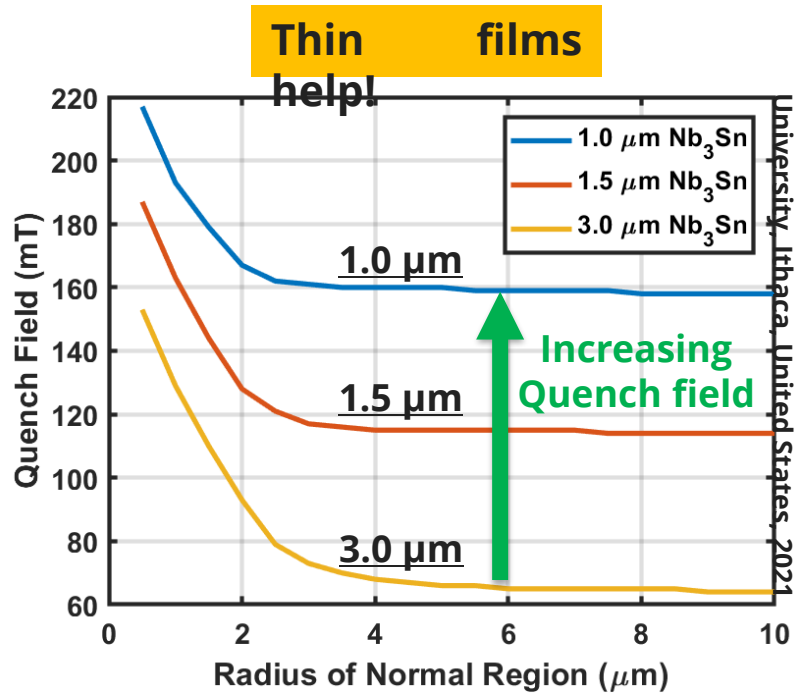
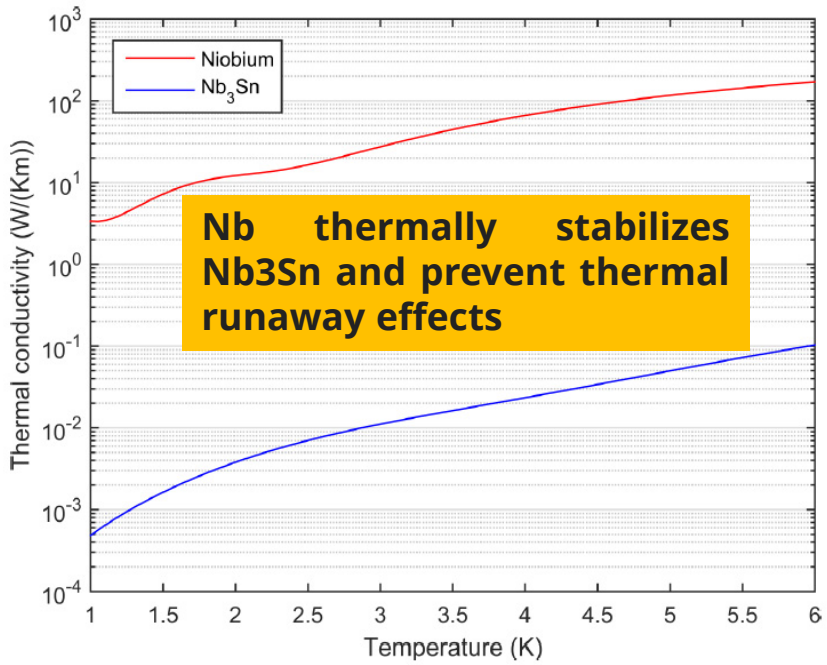




1. Thermal instability was identified as a major component of Nb_3Sn cavity quench.

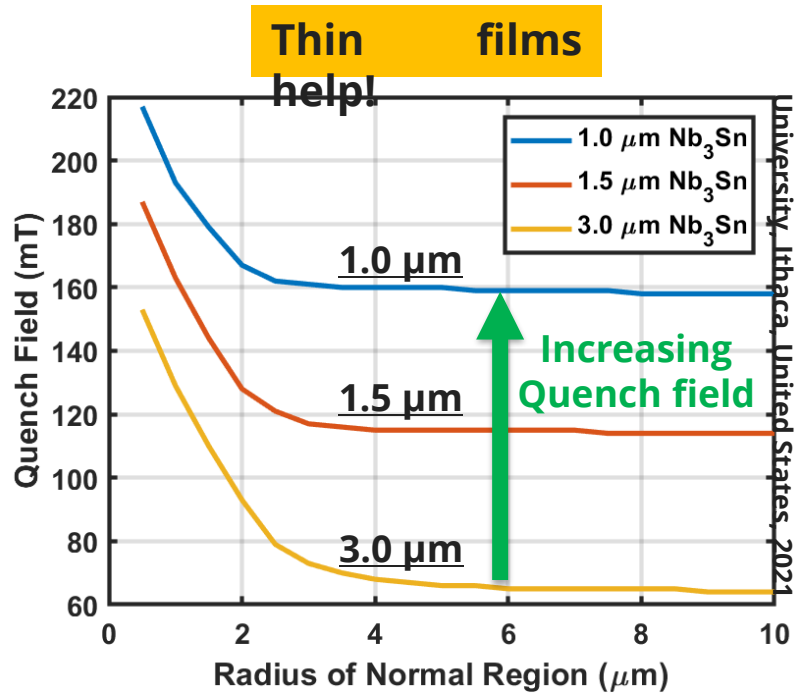
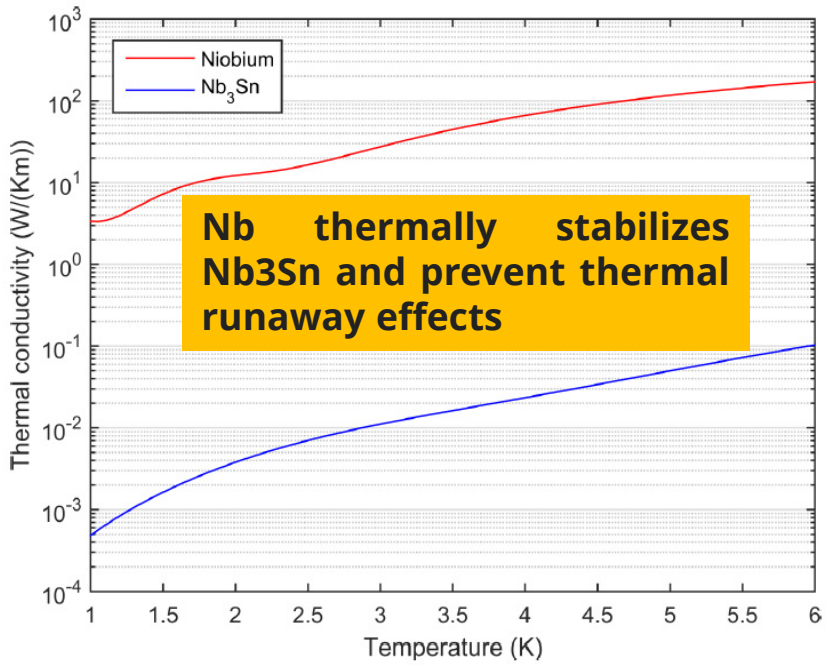


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R. D. Porter, Ph.D. thesis, Phys, Dept.,
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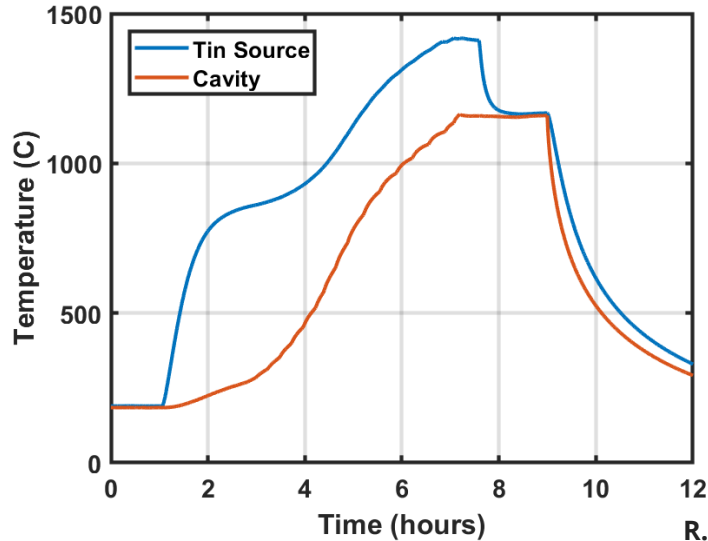
R. D. Porter, Ph.D. thesis, Phys, Dept.,
Cornell University, Ithaca, United States, 2021

2. Thinner films are smoother! (Pudasaini, WEIAA03)

The FNAL cavity which reached 24 MV/m had a shiny thin-film Nb₃Sn coating.
(S. Posen, <https://doi.org/10.1088/1361-6668/abc7f7>)

New recipe (inspired by FNAL) :

- **Increase Sn vapor pressure from start**
- **Shorter coating time**
- **Skip the 5-hour nucleation**



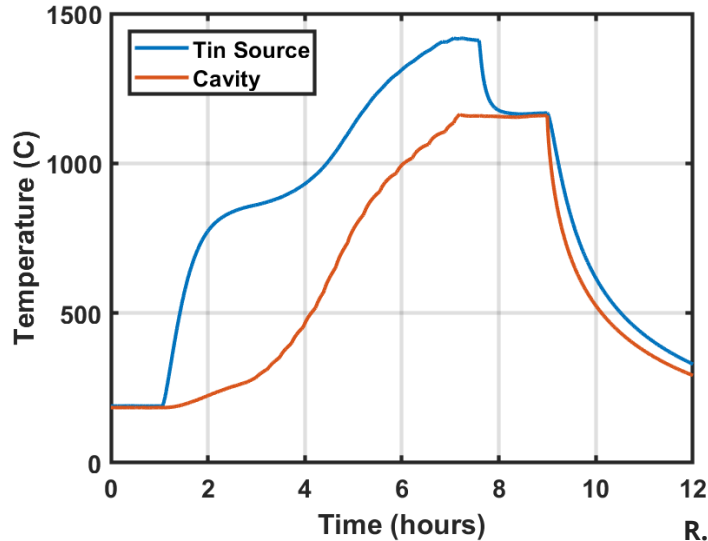
R. D. Porter, Ph.D. thesis, Phys, Dept.,
Cornell
University, Ithaca, United States, 2021

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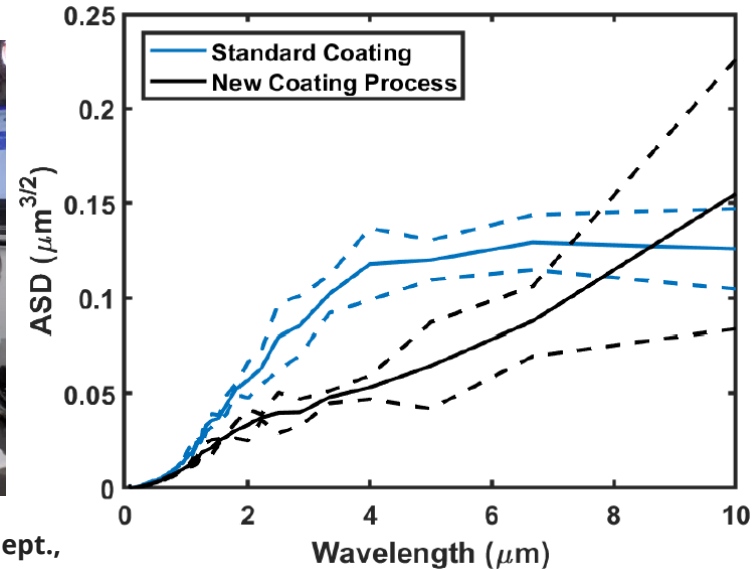
- Increase Sn vapor pressure from start
- Shorter coating time
- Skip the 5-hour nucleation

Result:

- Lower surface roughness!
- Lower BCS resistance, but higher R_0 and similar quench fields (suspect due to patchy regions)



R. D. Porter, Ph.D. thesis, Phys, Dept.,
Cornell
University, Ithaca, United States, 2021



New recipe:

→ Increase Sn vapor pressure from start

Result:

→ Lower surface roughness! ($R_a = 94\text{nm}$)
→ Lower BCS resistance but higher R_0

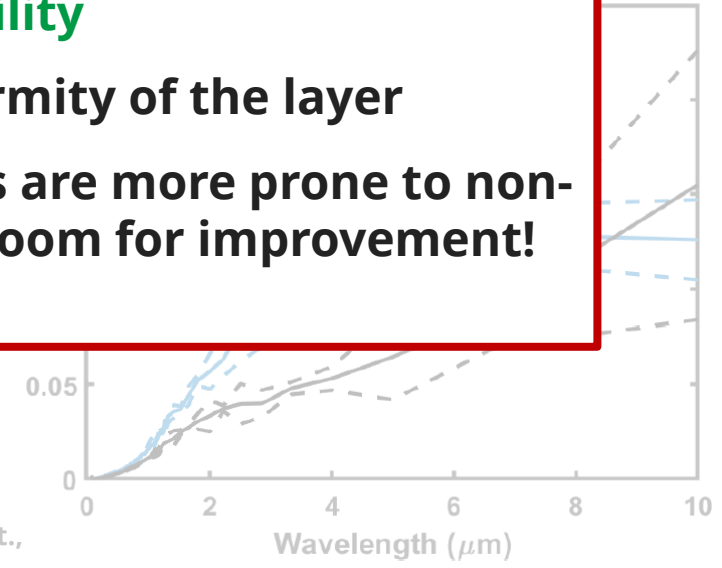
Thinner Nb_3Sn films show promise to achieve higher gradients due to **better thermal stability**

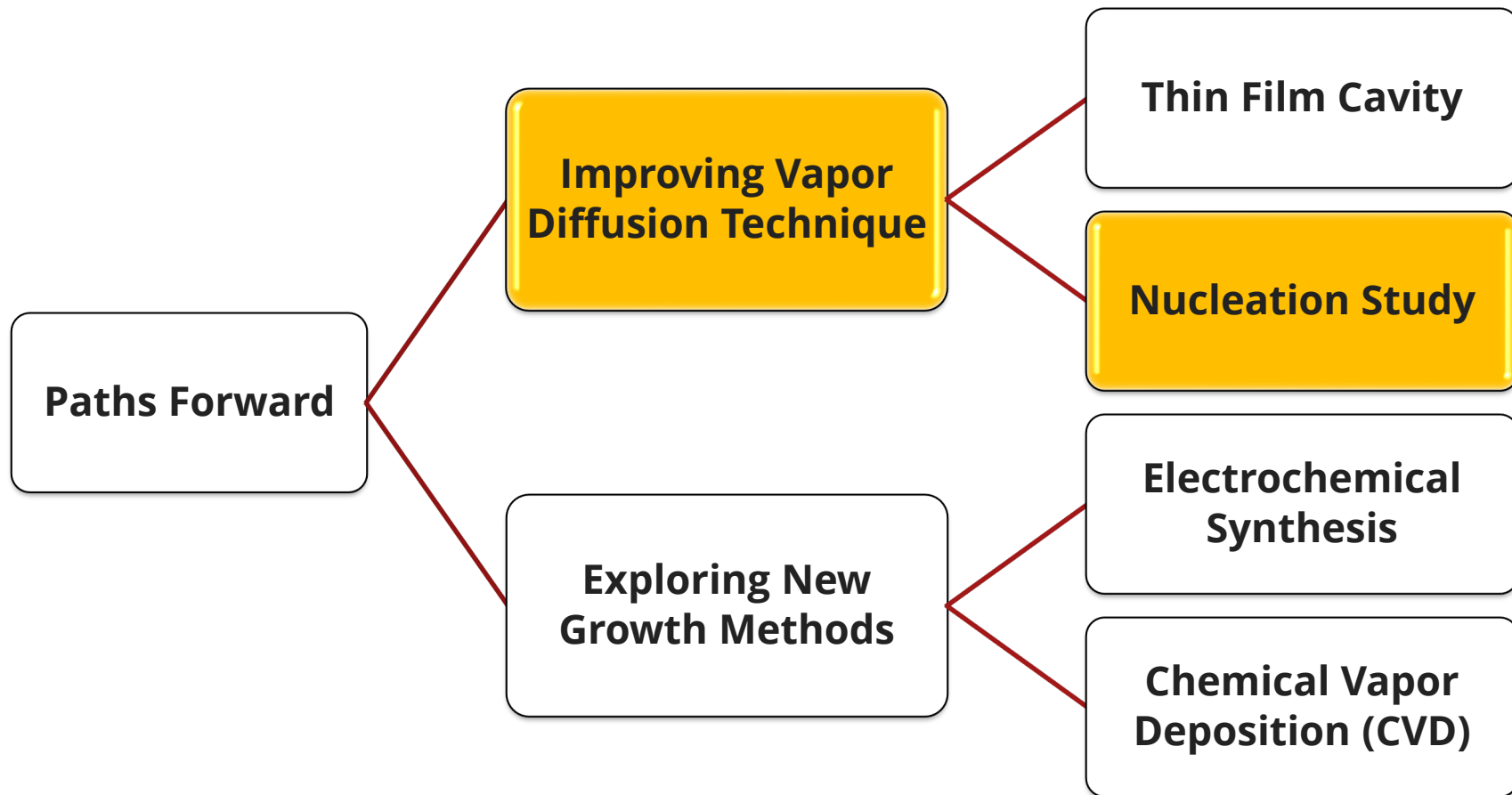
Key challenge: controlling the uniformity of the layer

→ **Reproducibility is an issue** as thin films are more prone to non-uniform patchy regions, but there is room for improvement!



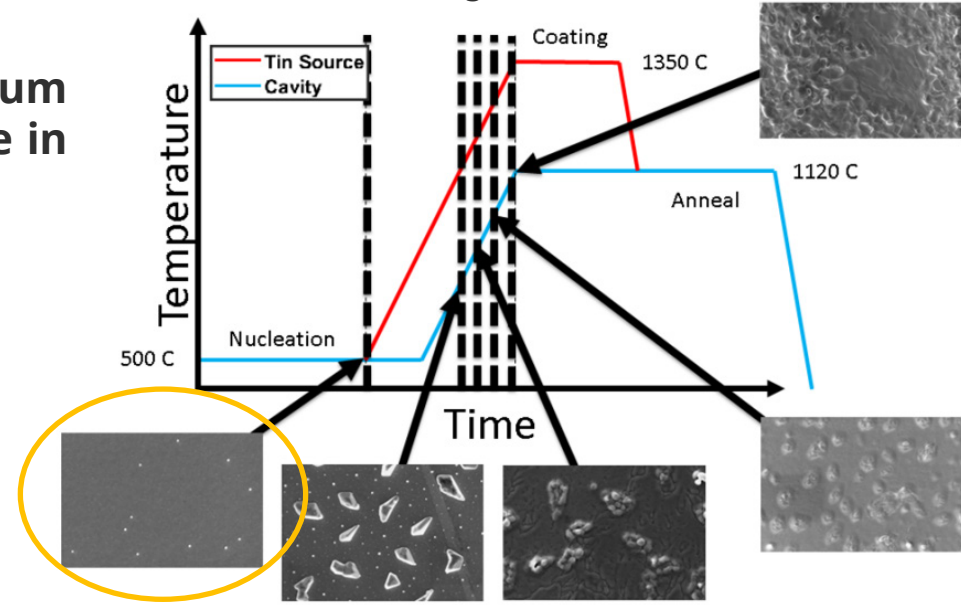
R. D. Porter, Ph.D. thesis, Phys, Dept., Cornell University, Ithaca, United States, 2021





Goal: Minimize tin depleted regions and achieve a smooth Nb_3Sn film.

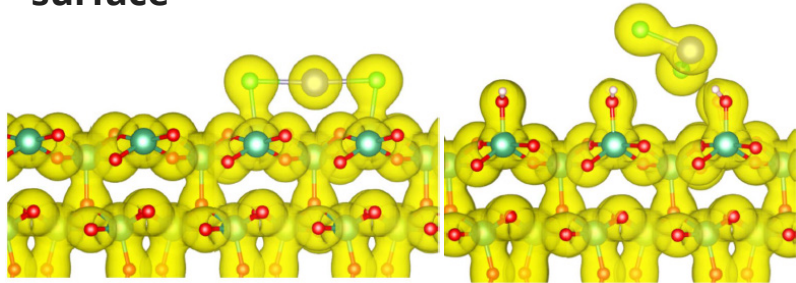
Motivation: The surface of the niobium oxide structure plays an important role in the nucleation of SnCl_2 .



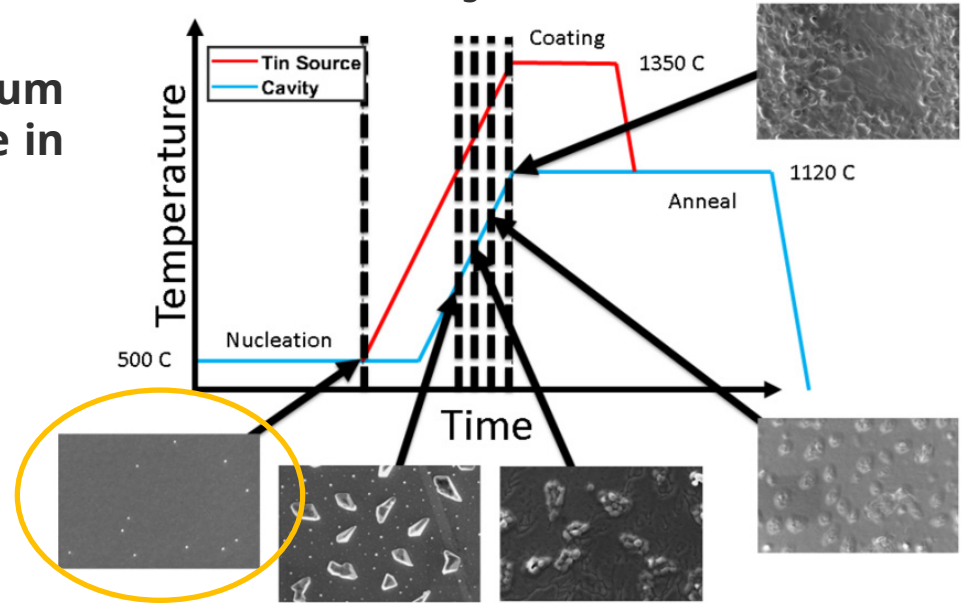
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SnCl_2 molecules with (right) and without (left) passivating OH groups covering the surface



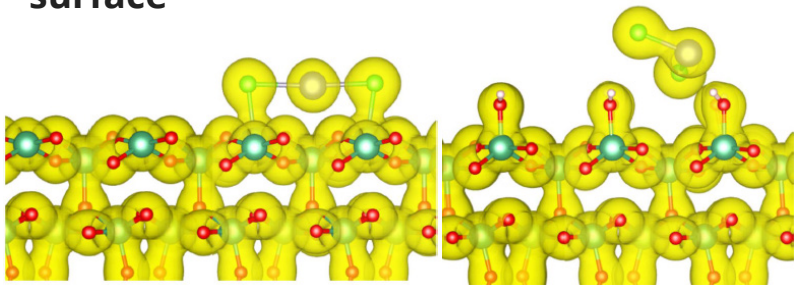
N. Sitaraman, "Theory work on SRF materials", Ph.D. thesis, Phys, Dept., Cornell University, Ithaca, United States, 2022



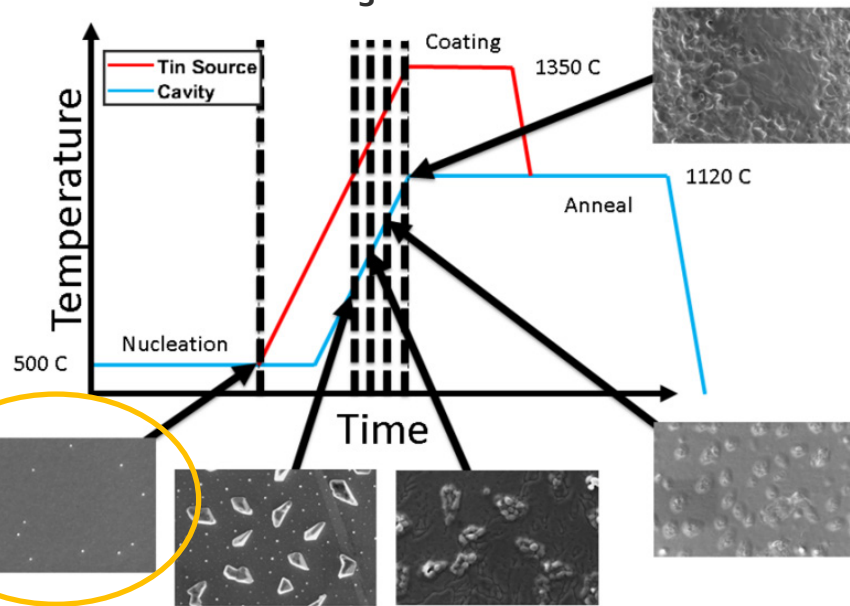
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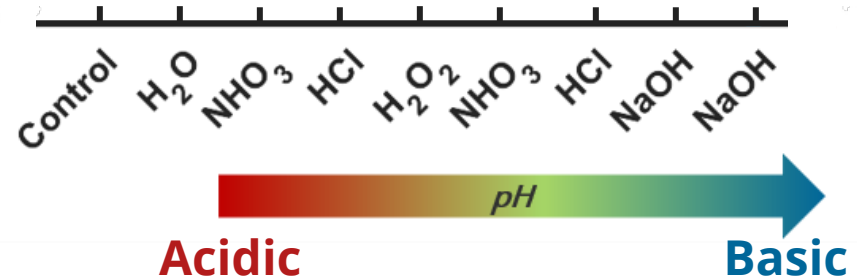
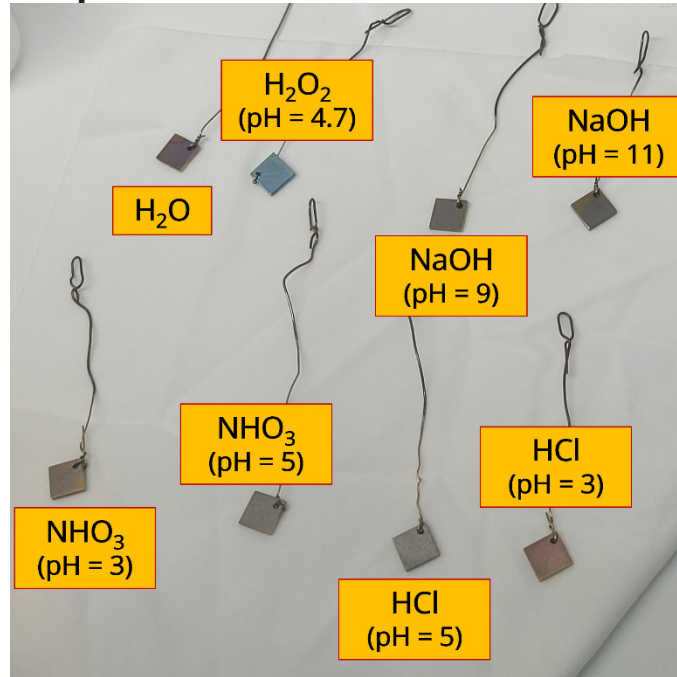


N. Sitaraman, "Theory work on SRF materials", Ph.D. thesis, Phys, Dept., Cornell University, Ithaca, United States, 2022



Can we experimentally promote more uniform nucleation through adding or removing OH groups from the oxide?

Samples were electropolished, anodized and soaked in chemical treatments in a nitrogen atmosphere.

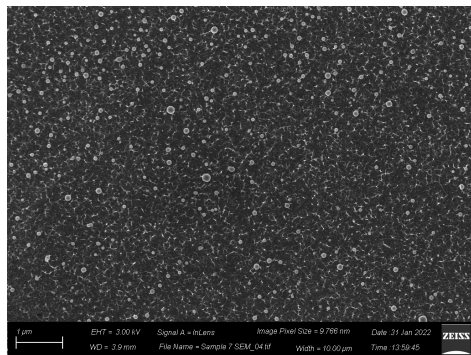


The furnace run was stopped after the nucleation step and the samples were taken out for surface characterization and analysis.

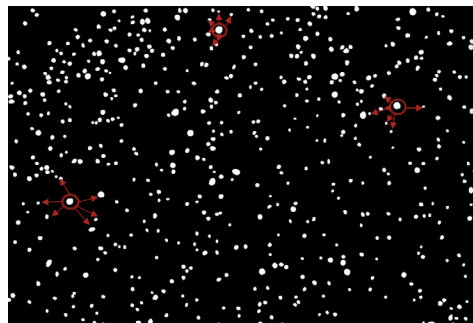
The samples (control sample not shown) after being soaked in the chemical treatments.

1. Scanning Electron Microscopy (SEM)

→ extract information about distribution of nucleation sites (*"droplets"*)



Post-processing

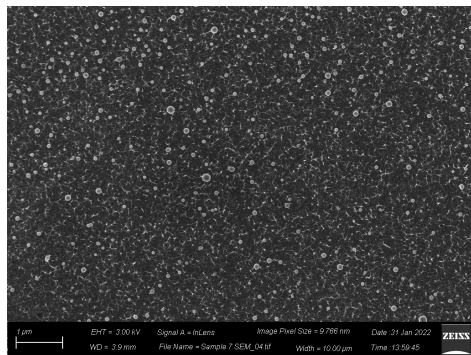


Density

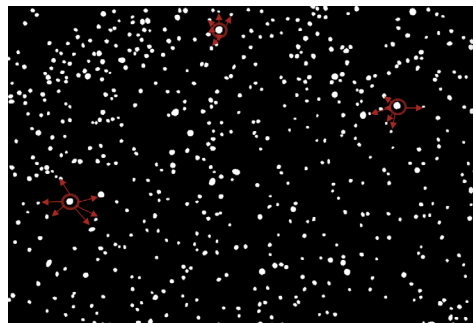
Avg. distance to six nearest neighbors

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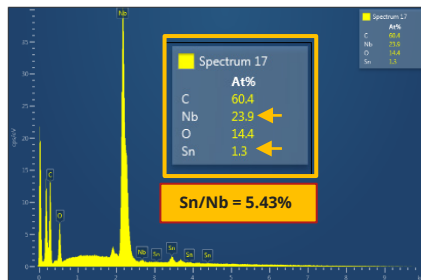
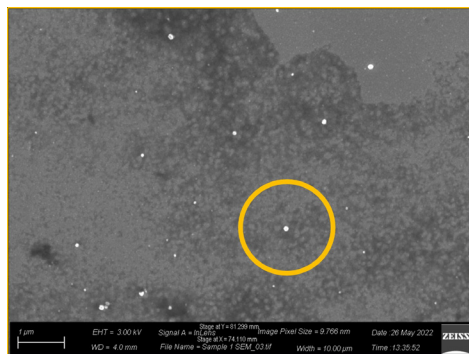


Density

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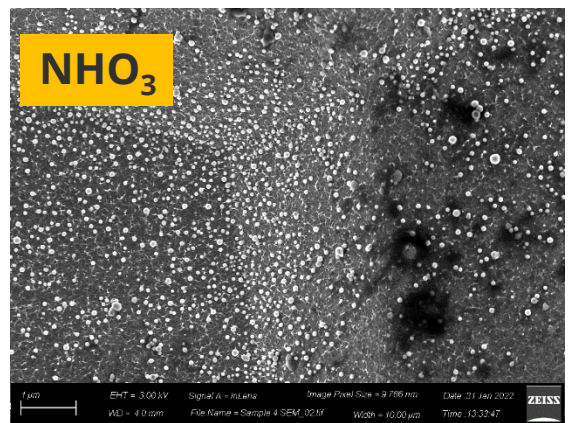
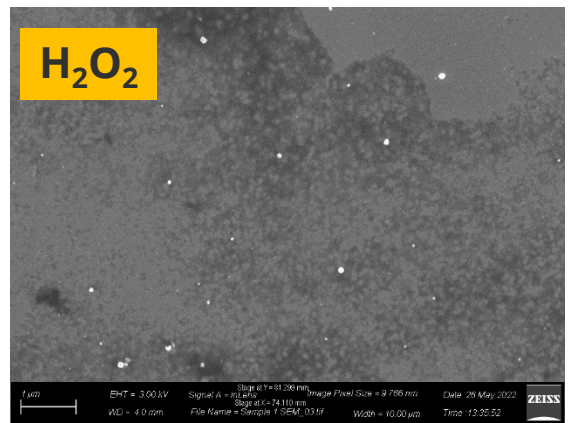
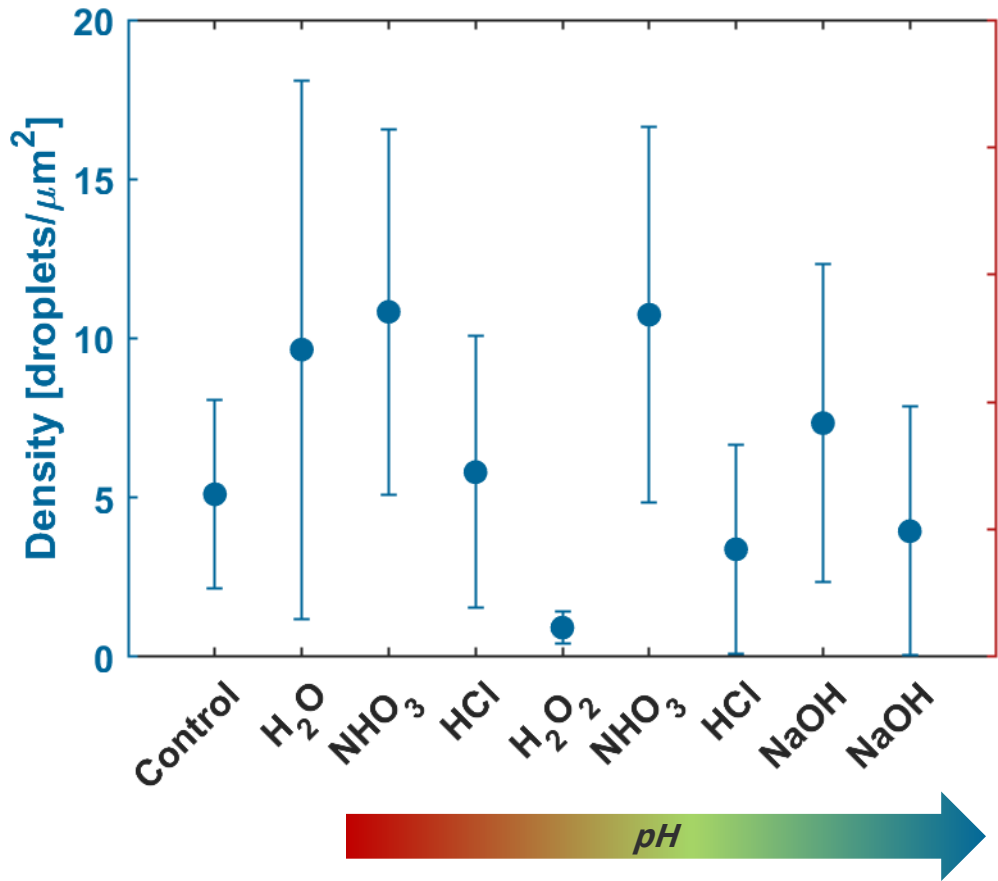
2. Energy Dispersive Spectroscopy (EDS)

→ extract information about elemental composition (Sn:Nb atomic

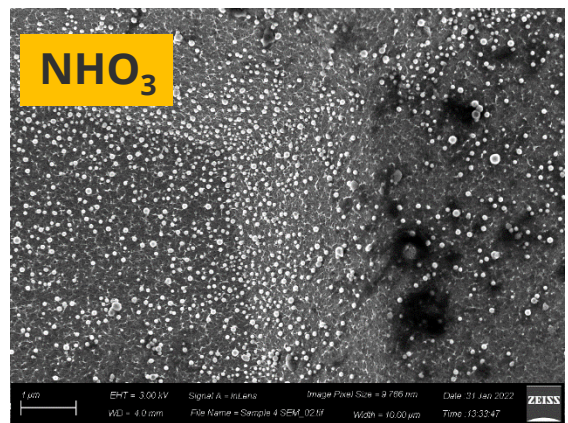
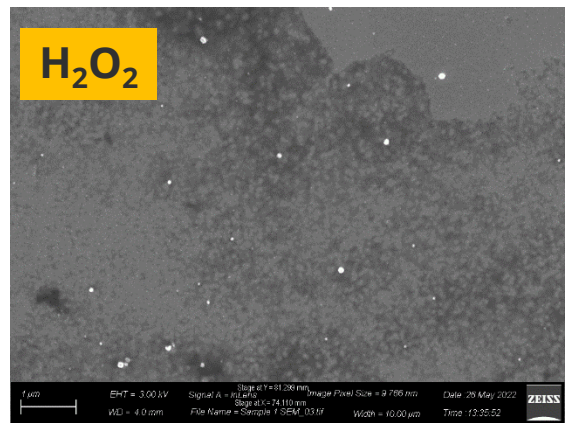
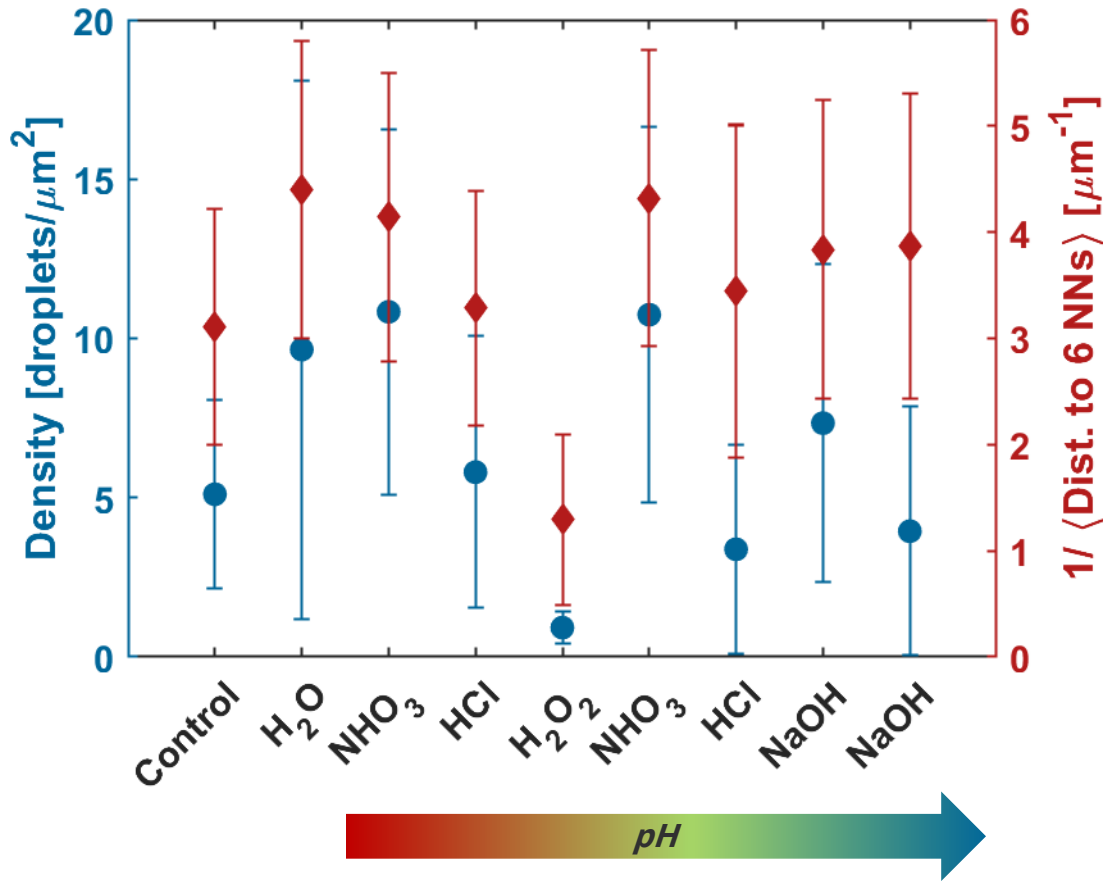


Establish **relative comparison** of the **Sn:Nb at. % ratio** among our samples.

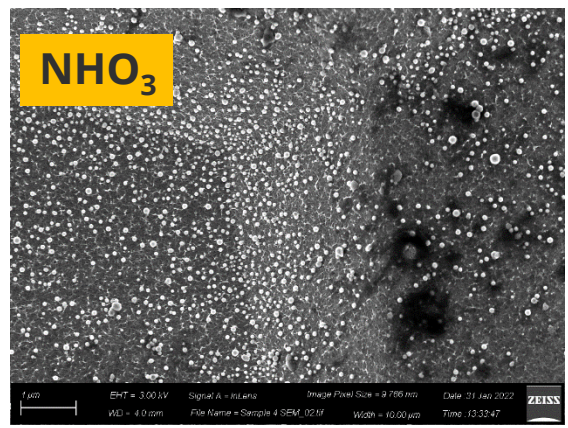
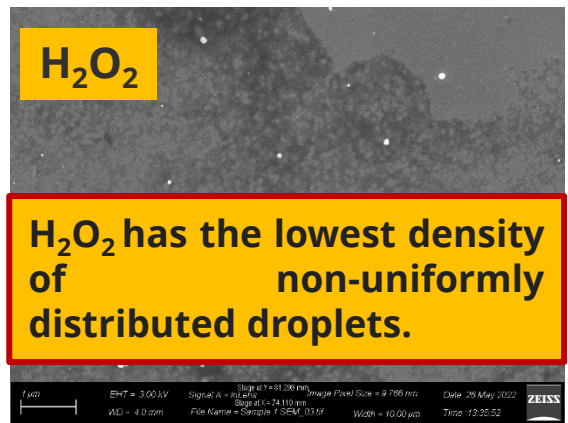
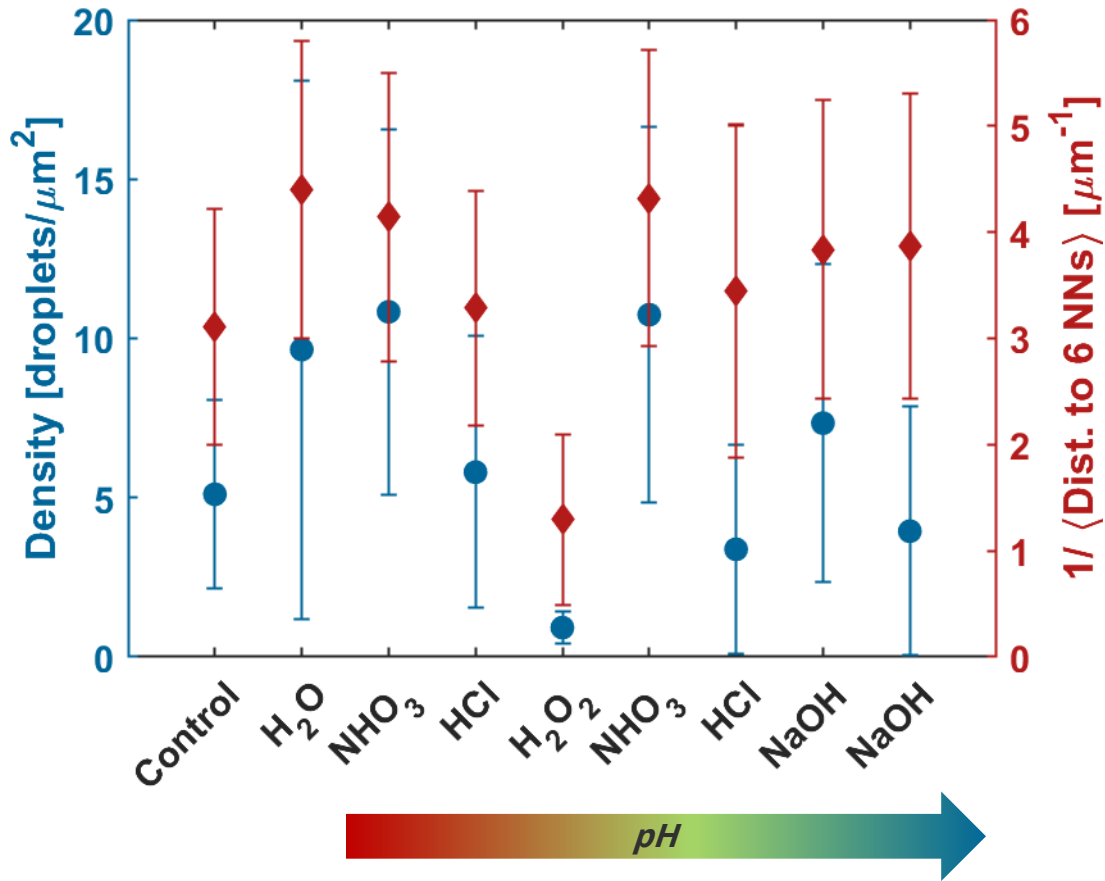
Distribution of Nucleation Sites



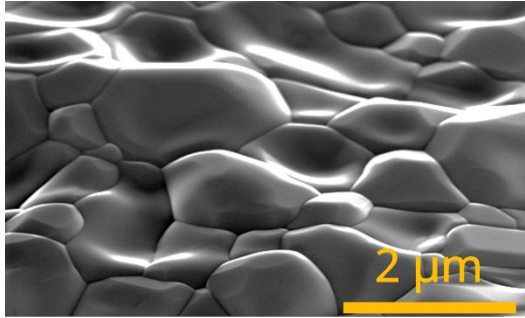
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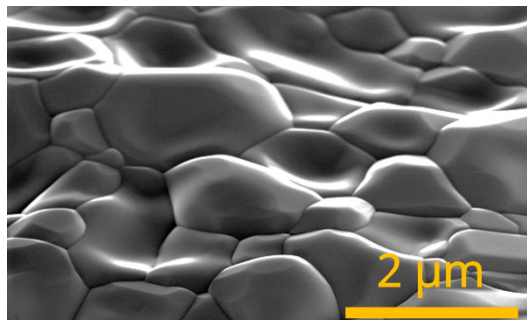
Nb_3Sn grains have $\sim 1 \mu\text{m}^2$ area.



We want at least 1 nucleation site per intended grain:

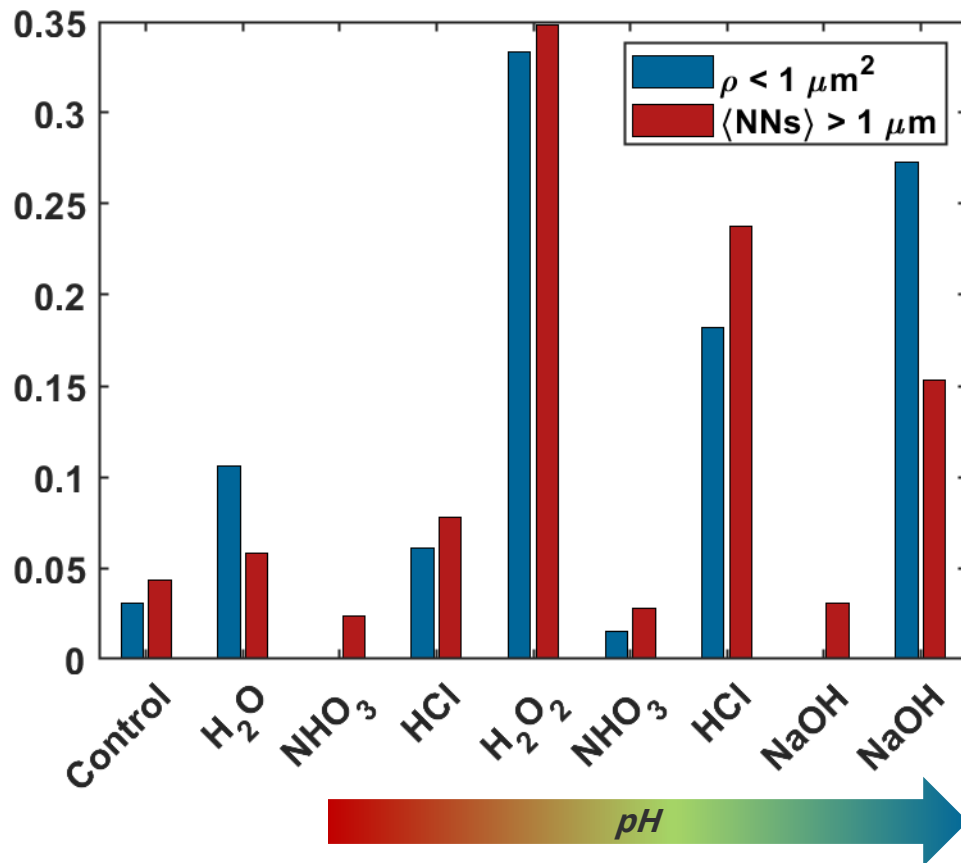
1. How many imaged areas had a **density of less than one particle per $1 \mu\text{m}^2$ area?**
2. How many **avg. distances to 6 nearest neighbors exceed $1 \mu\text{m}$?**

Nb₃Sn grains have ~1 μm² area.

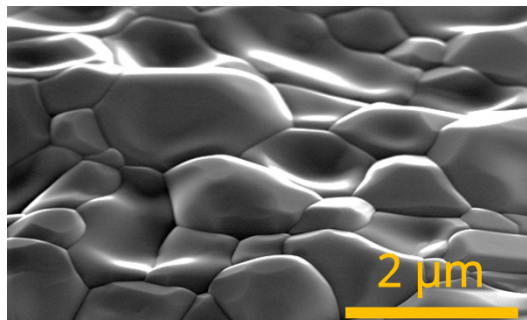


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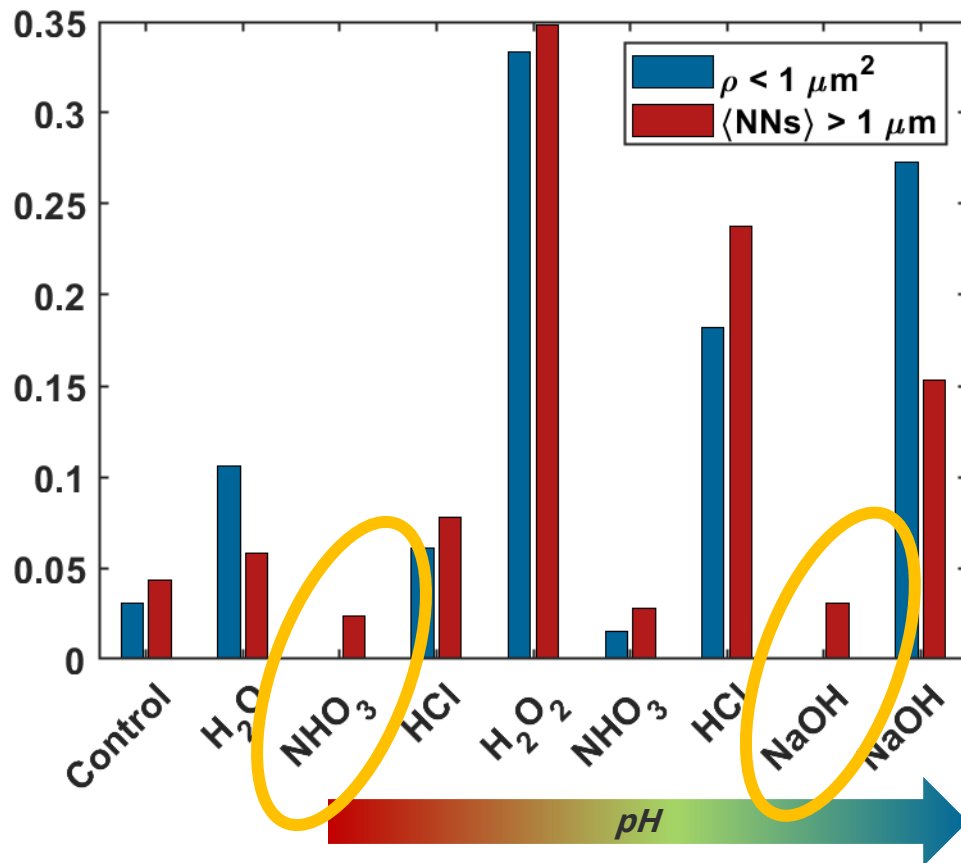


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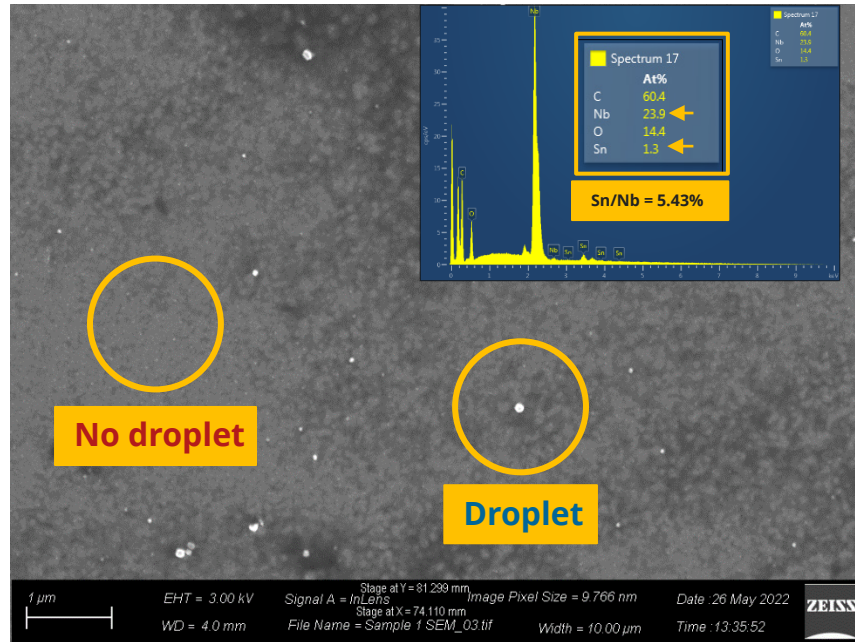


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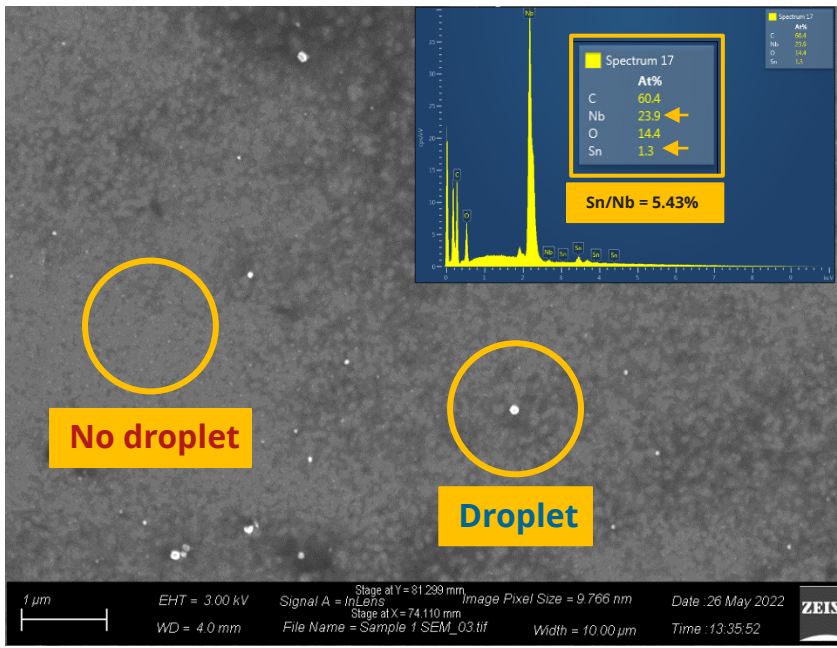
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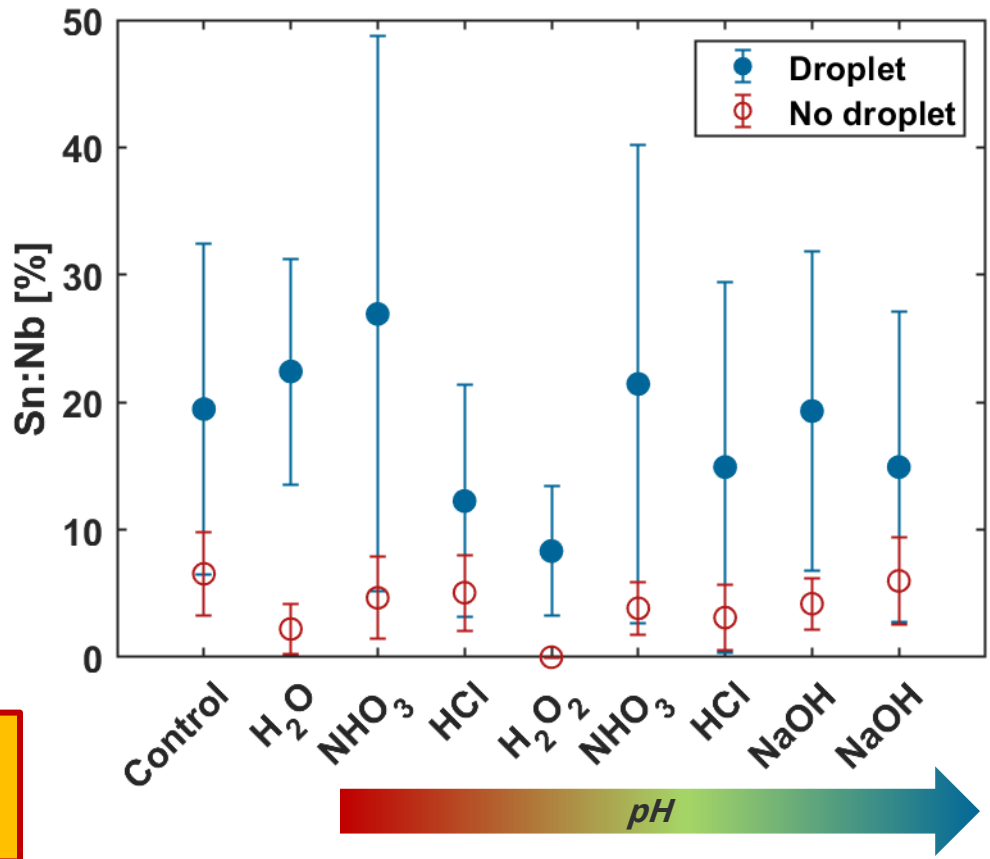
Are these droplets actually tin?



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All samples (except H₂O₂) have tin present even when no droplets are seen in the SEM image!

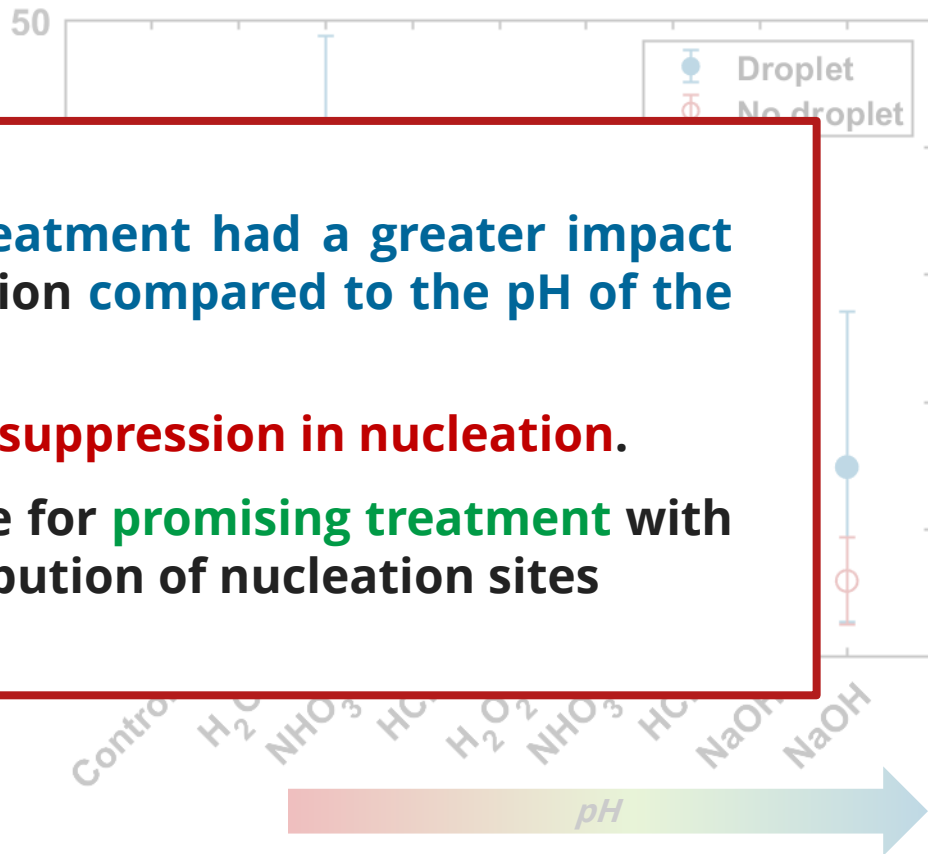


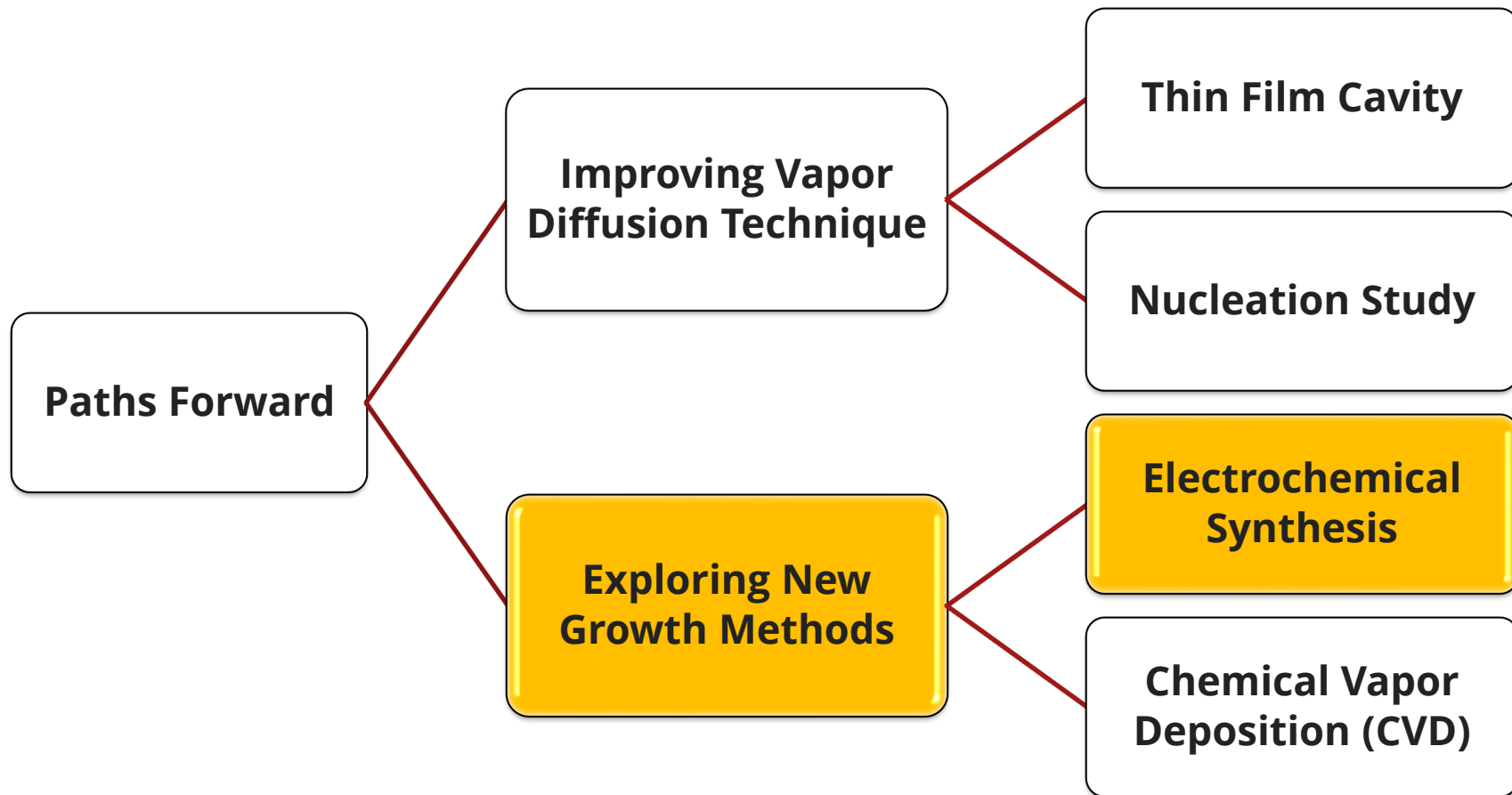
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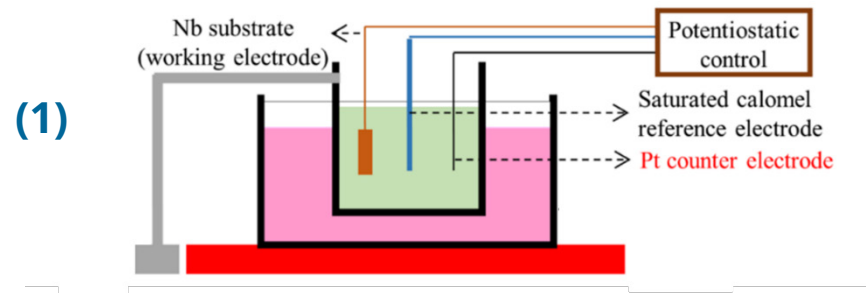


- The type of chemical treatment had a greater impact on the quality of nucleation compared to the pH of the treatment.
- H_2O_2 shows a significant suppression in nucleation.
- NHO_3 is a good candidate for promising treatment with dense and uniform distribution of nucleation sites

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Pre-deposition of Sn onto Nb

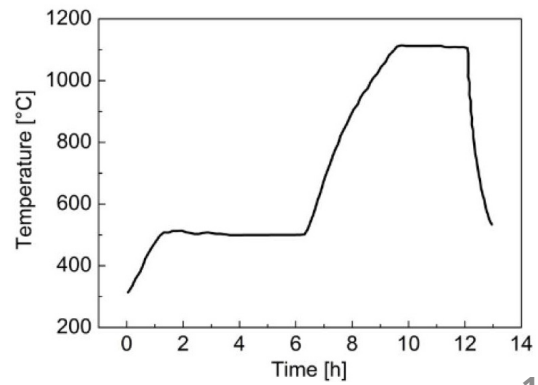


Advantages:

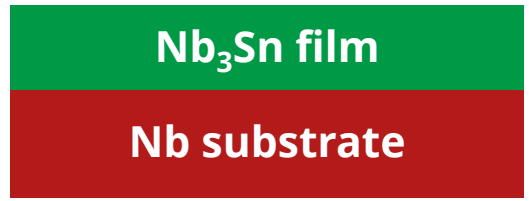
- Promotes **uniform** distribution of nucleation (↓ **surface roughness**)
- Provides **sufficient Sn supply** in critical times



Post Annealing



(2)



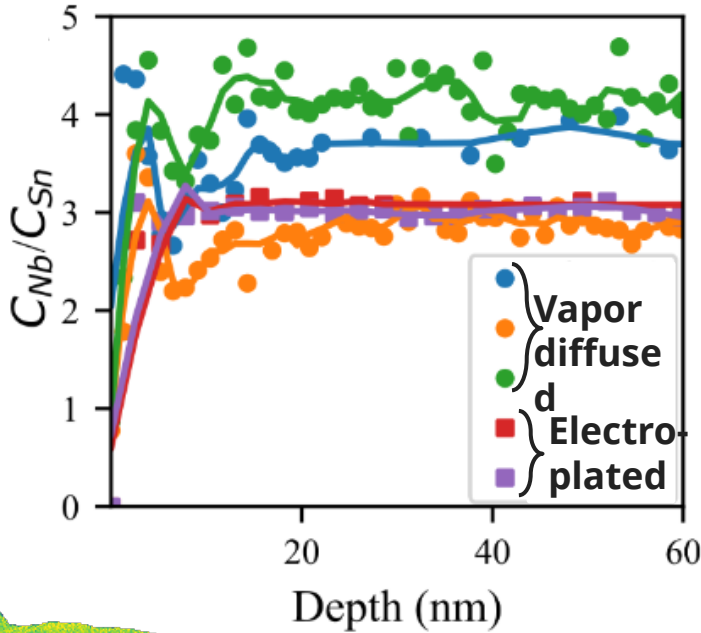
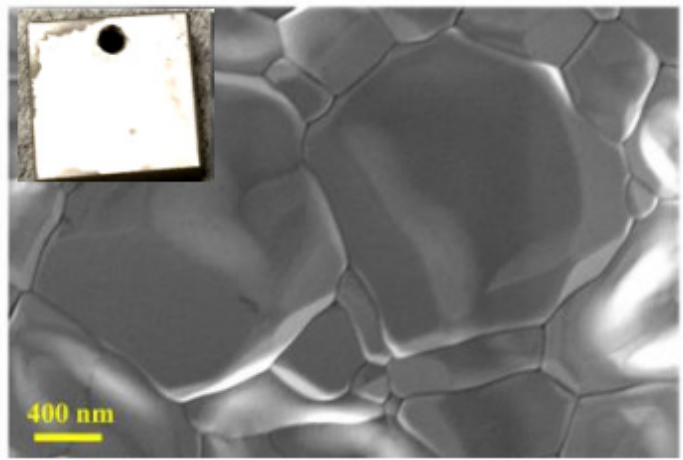
Conversion to stoichiometric Nb₃Sn



This method was developed by Zeming Sun Sun et al.,

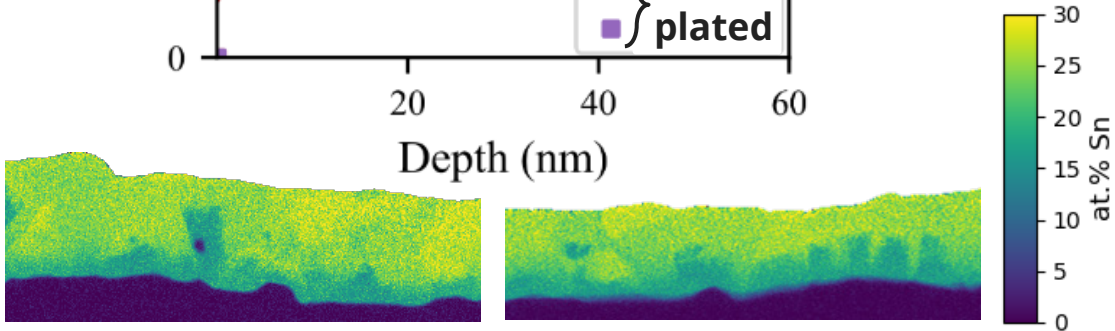
<https://doi.org/10.48550/arXiv.2302.02054>

Sample studies have shown low surface roughness and little variation in Sn concentration with depth.



This method was developed by Zeming Sun

Sun et al., doi: 10.48550/arXiv.2302.02054

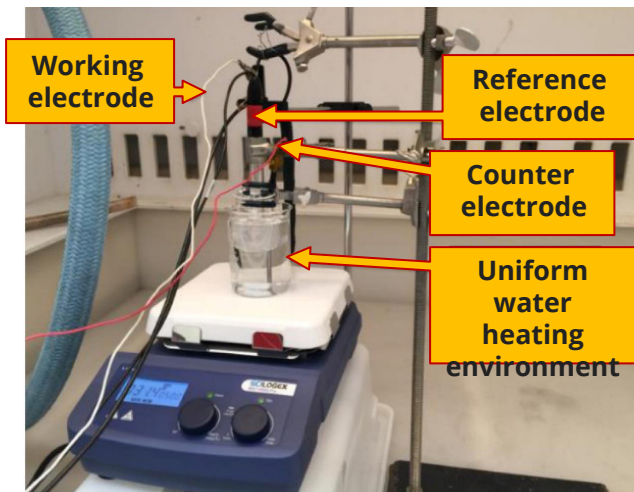


Vapor diffused Nb₃Sn

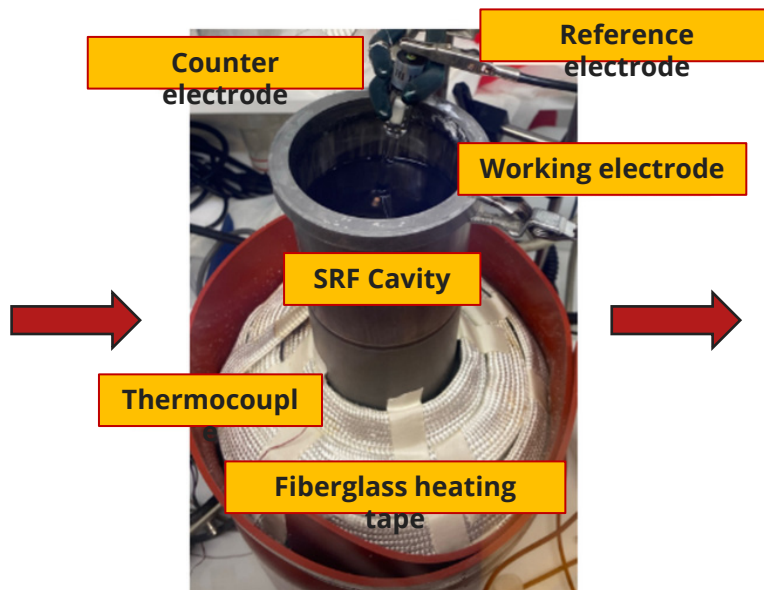
Electroplated Nb₃Sn

Images from Zhaslan Baraissov

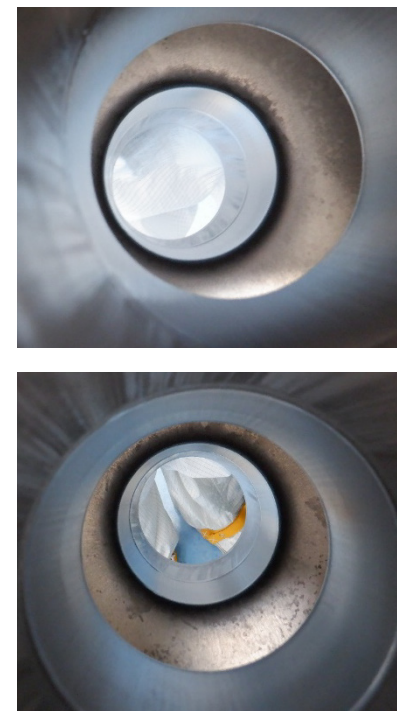
Samples



1.3 GHz cavity

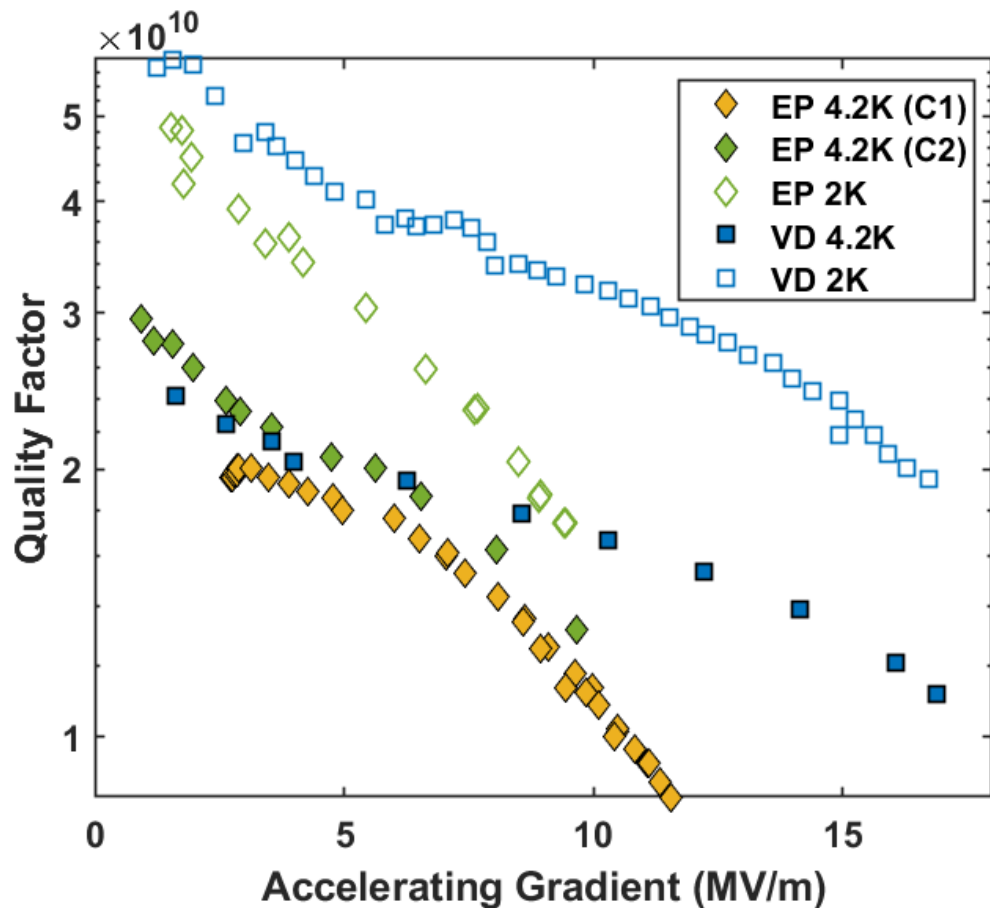


Post-annealing



This method was developed by Zeming Sun
Sun et al.,

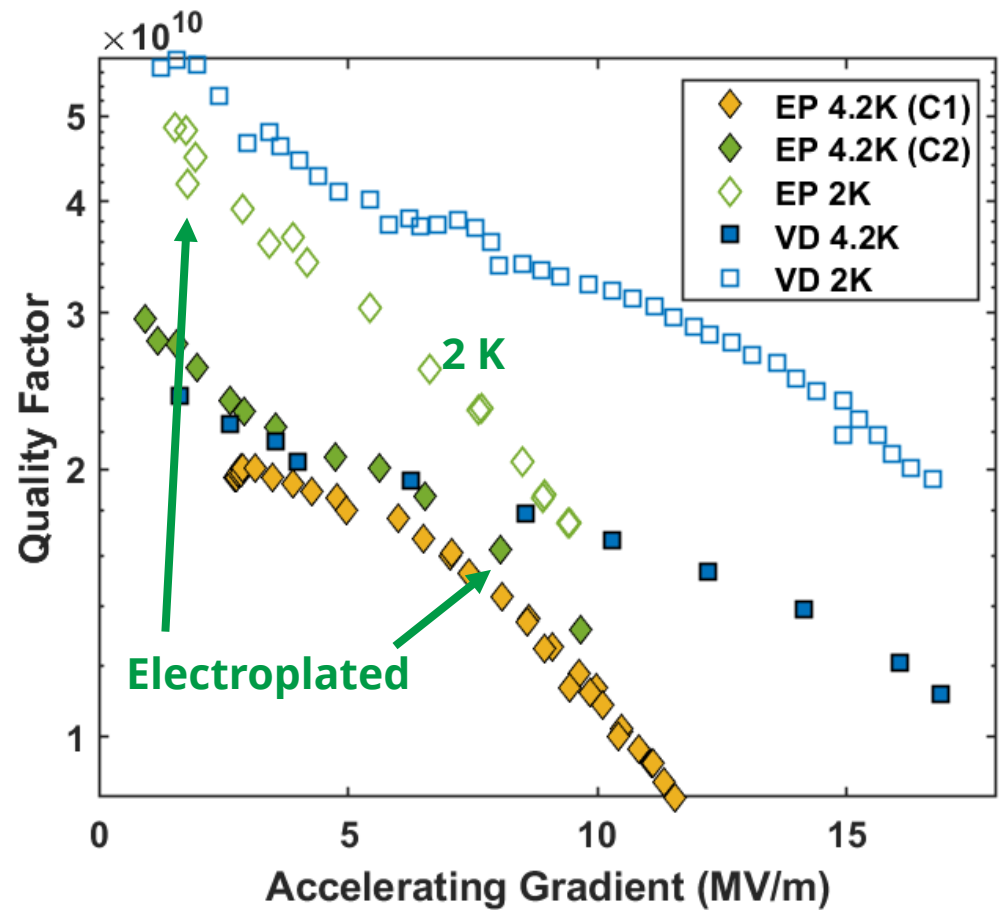
<https://doi.org/10.48550/arXiv.2302.02054>



Q_0 at 4.2K $> 10^{10}$

Quench field was ~ 13 MV/m

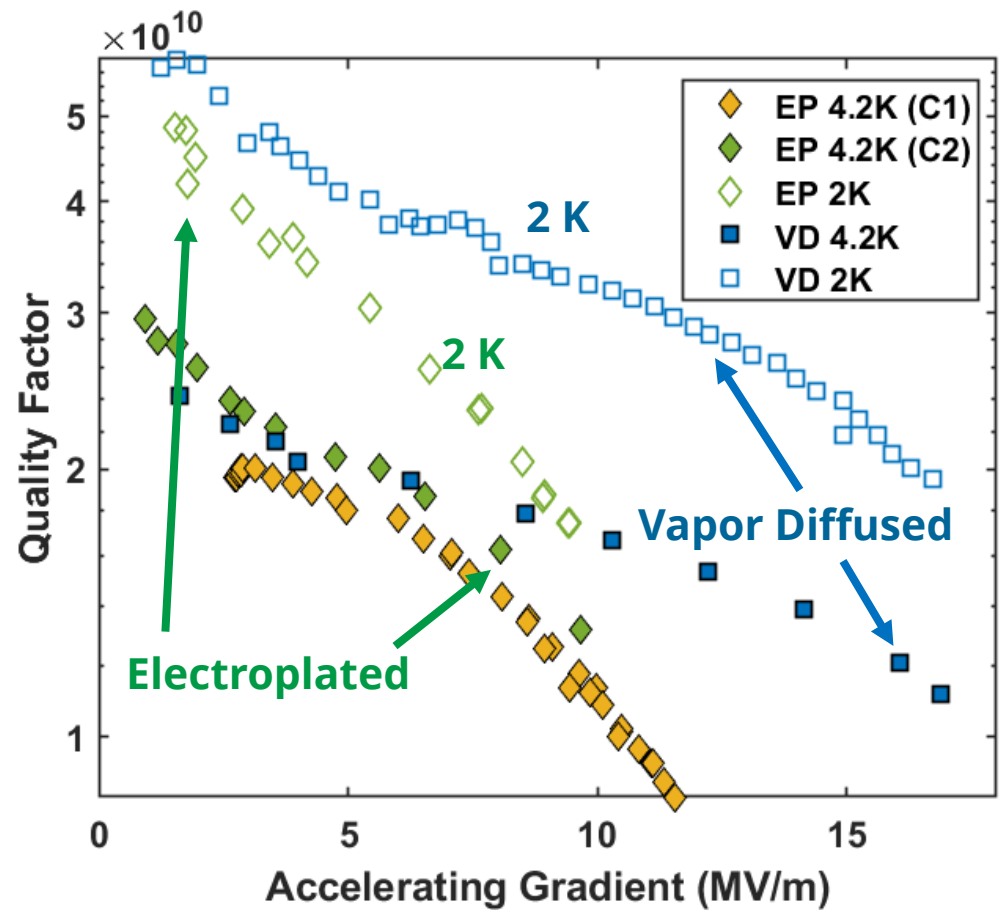
This is the first-ever alternative growth method of Nb₃Sn to achieve such high performance!



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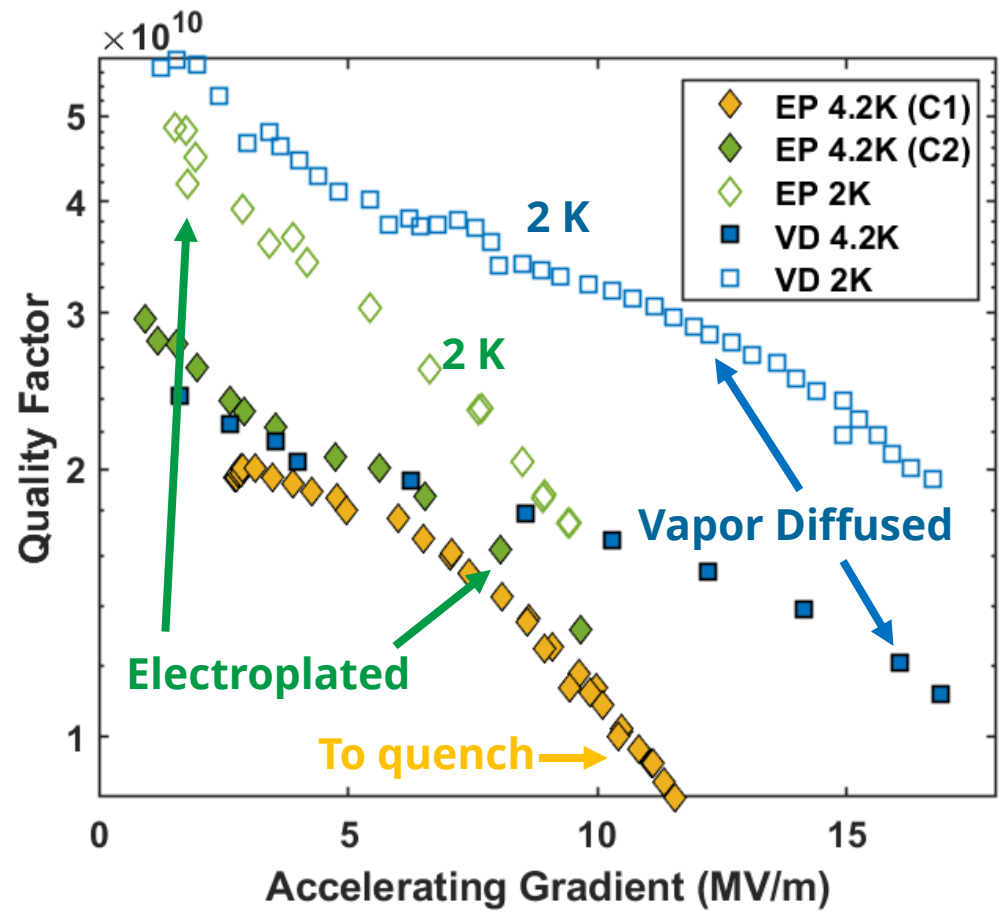
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Q_0 at 4.2K $> 10^{10}$

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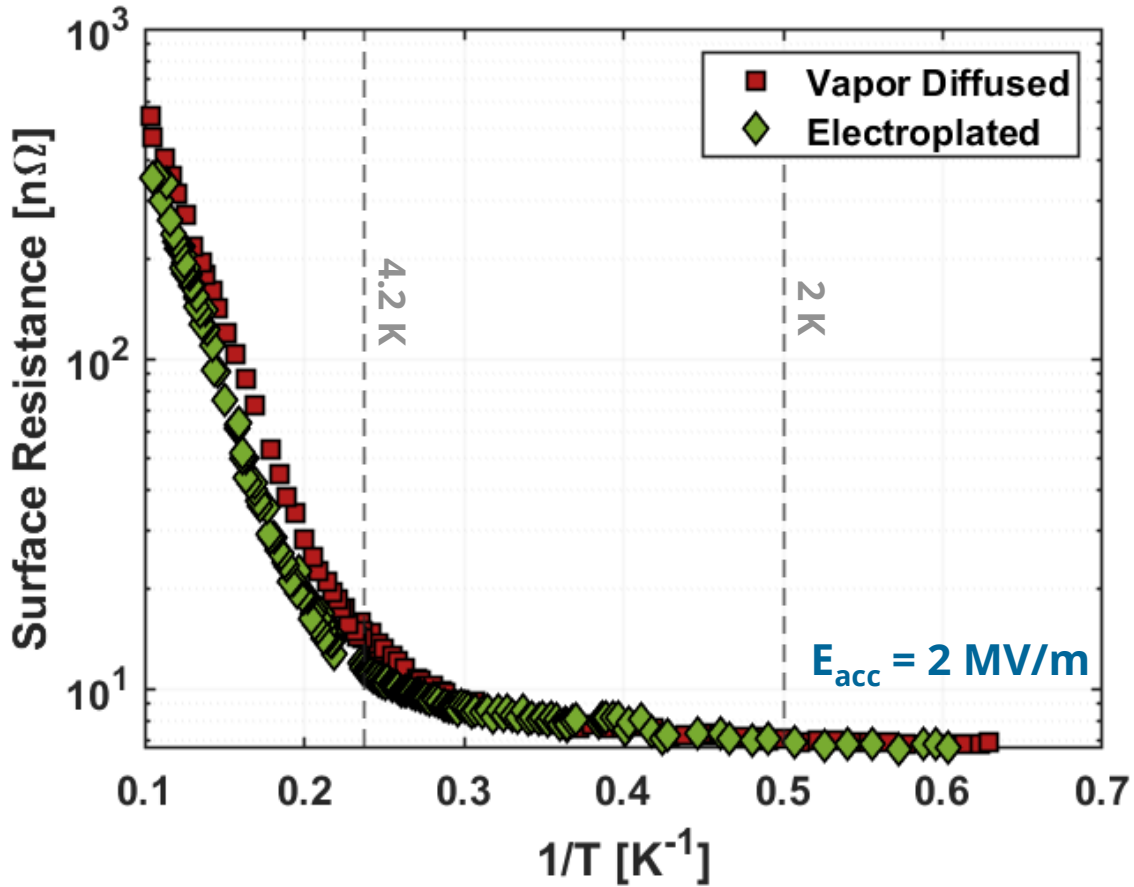
This is the first-ever alternative growth method of Nb₃Sn to achieve such high performance!

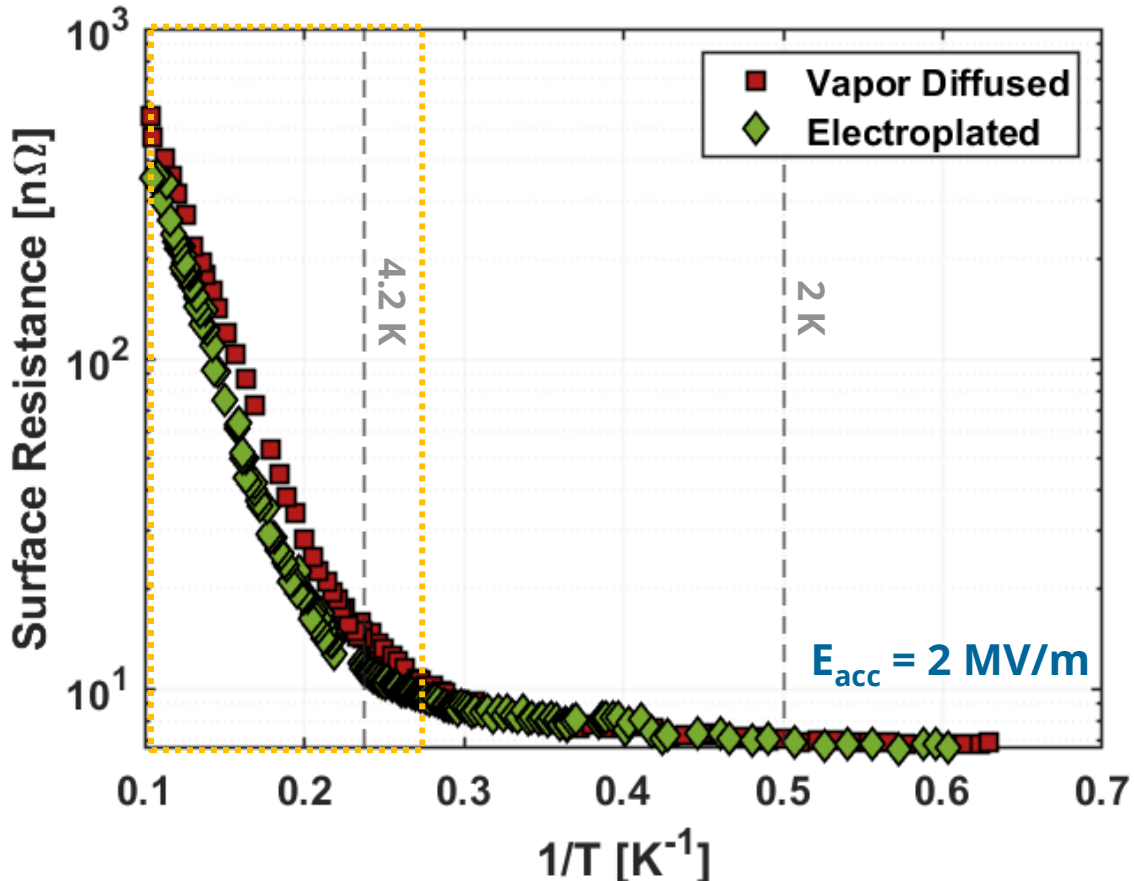


Q_0 at 4.2K $> 10^{10}$

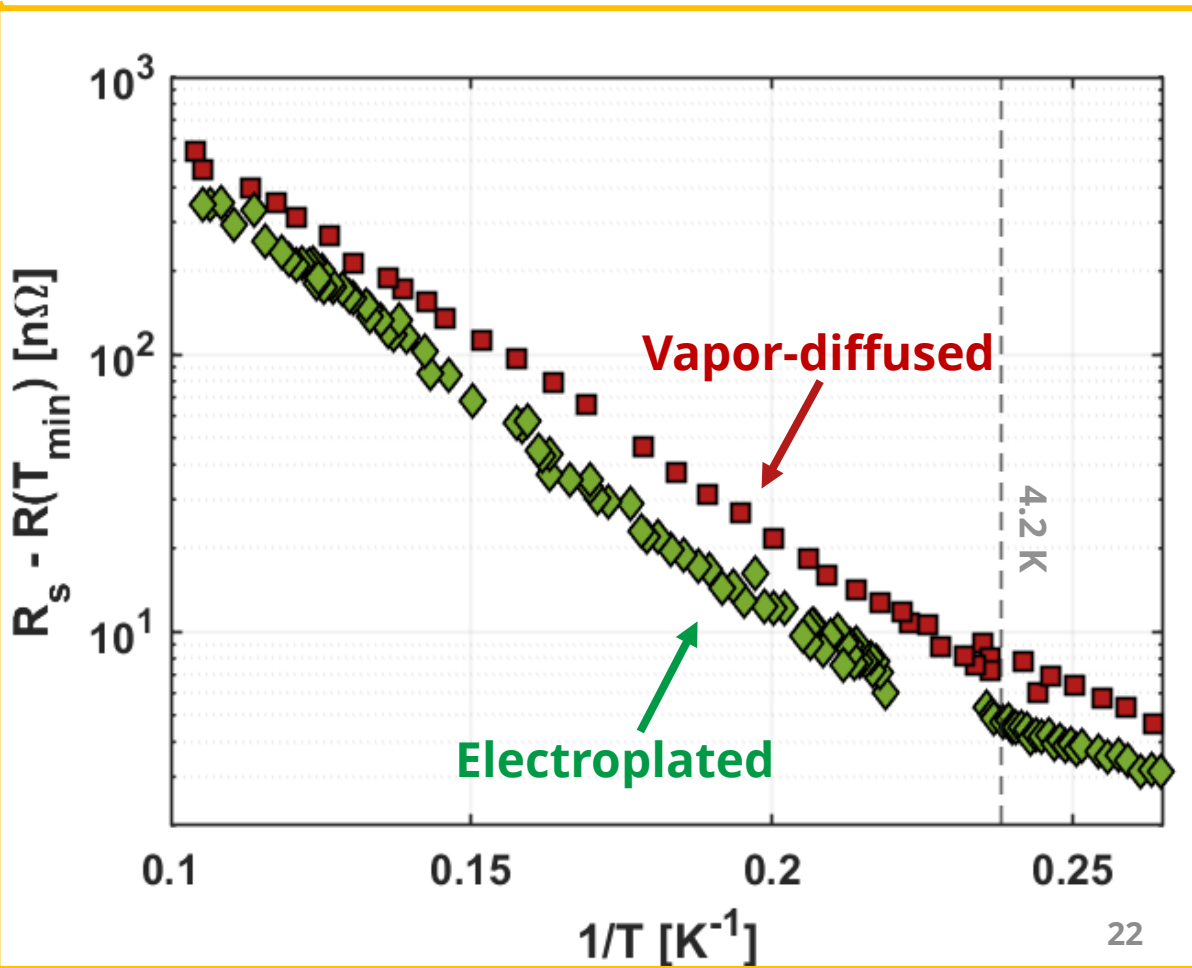
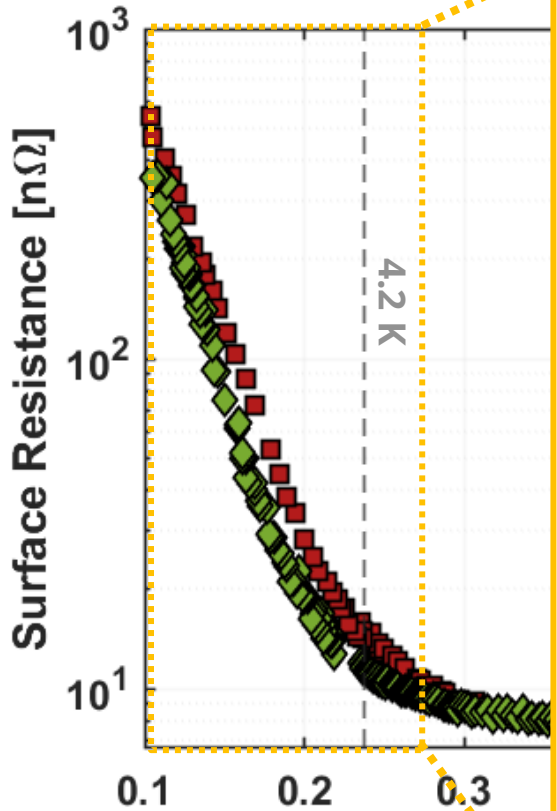
Quench field was ~ 13 MV/m

This is the first-ever alternative growth method of Nb₃Sn to achieve such high performance!

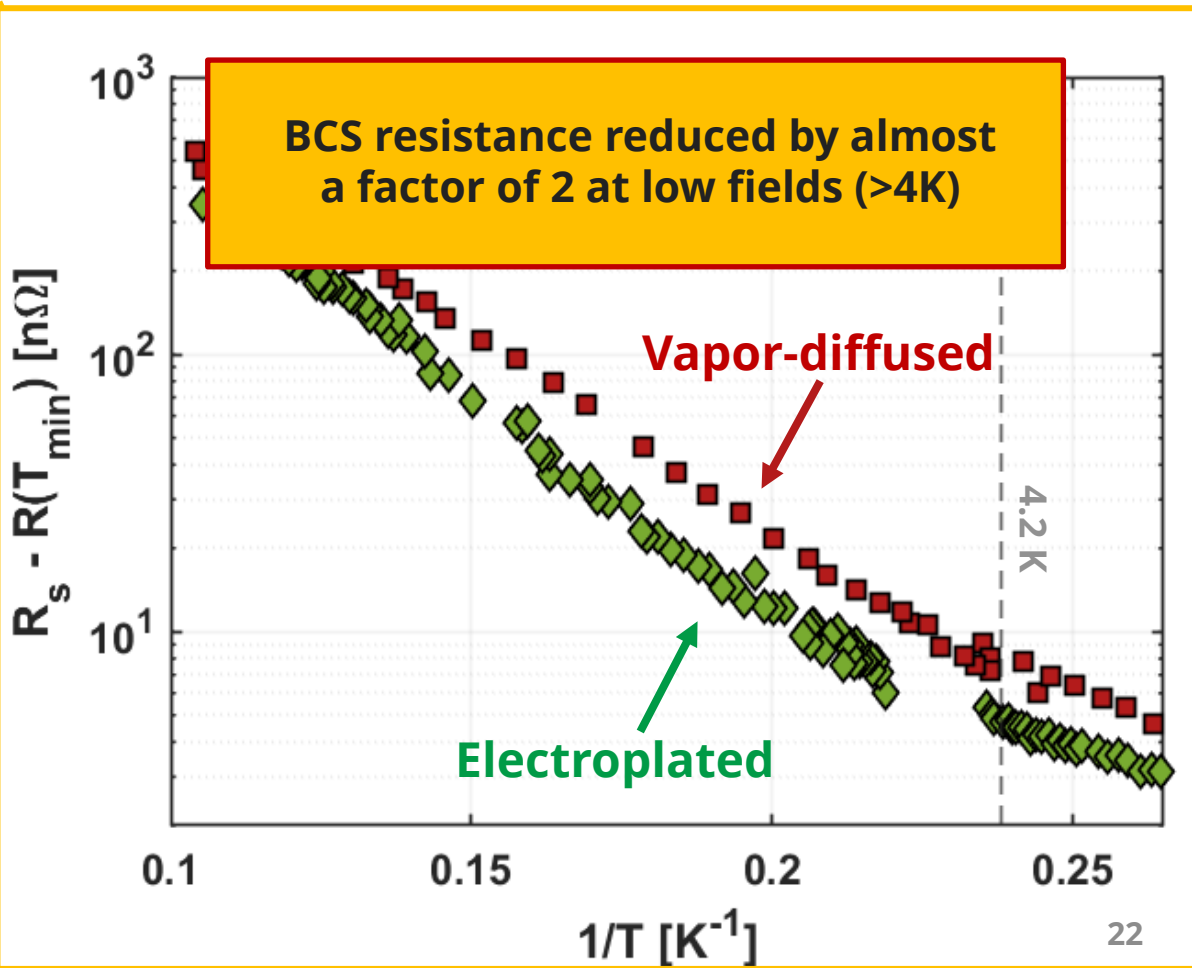
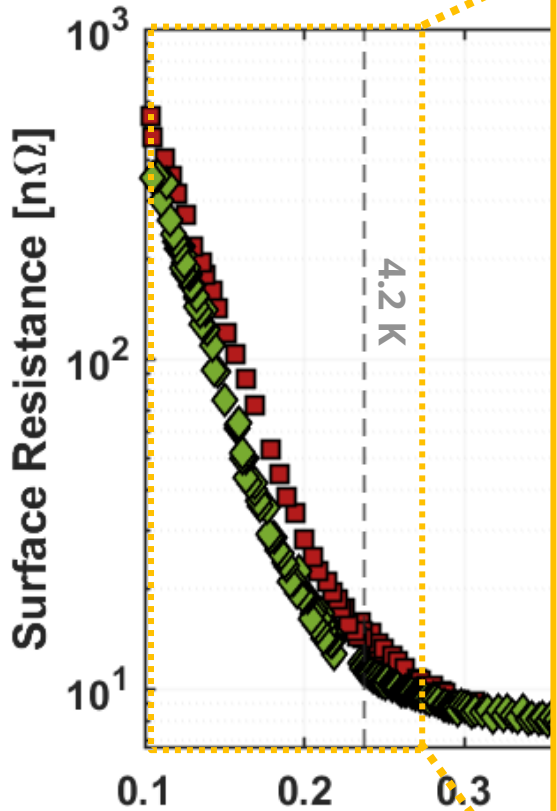


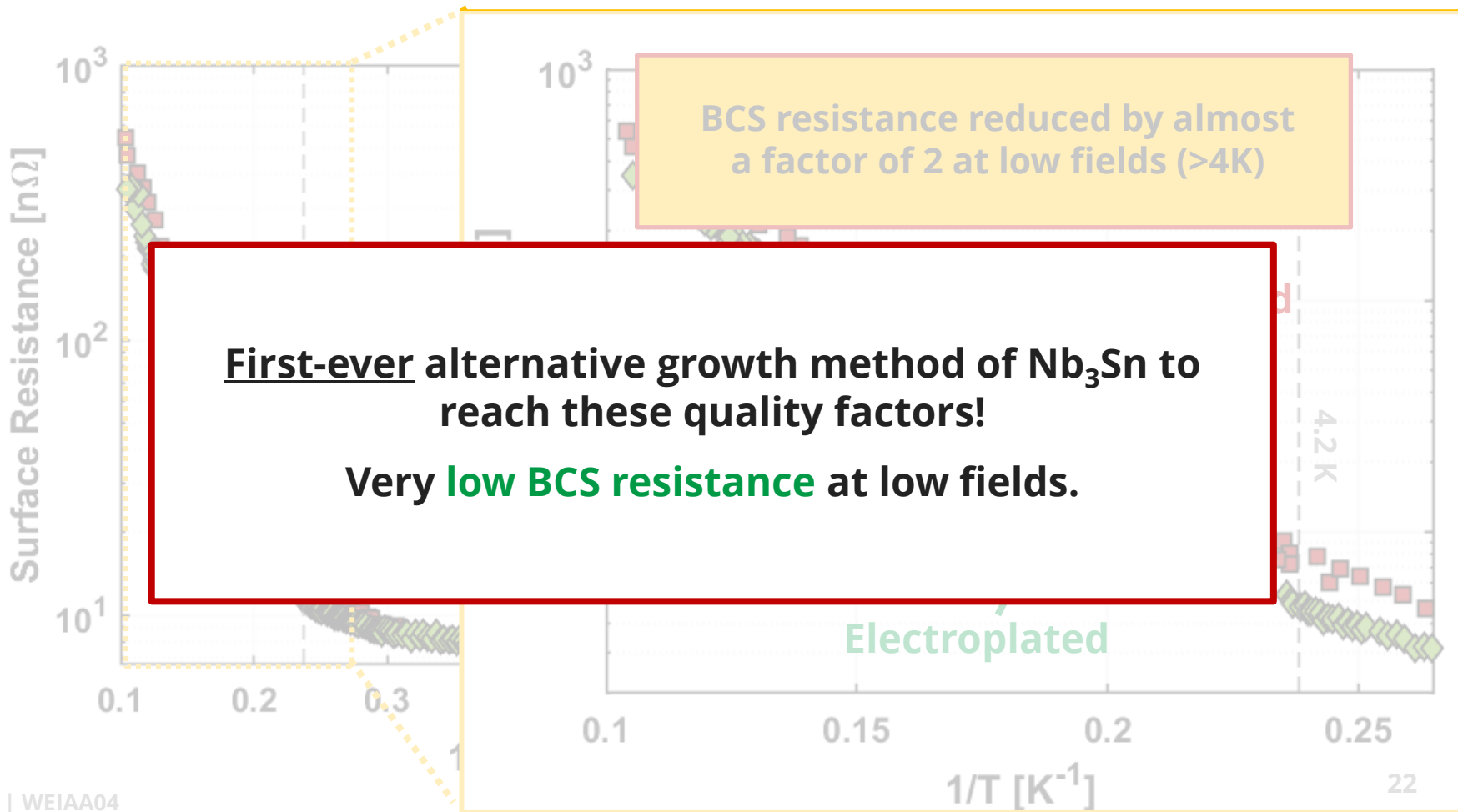


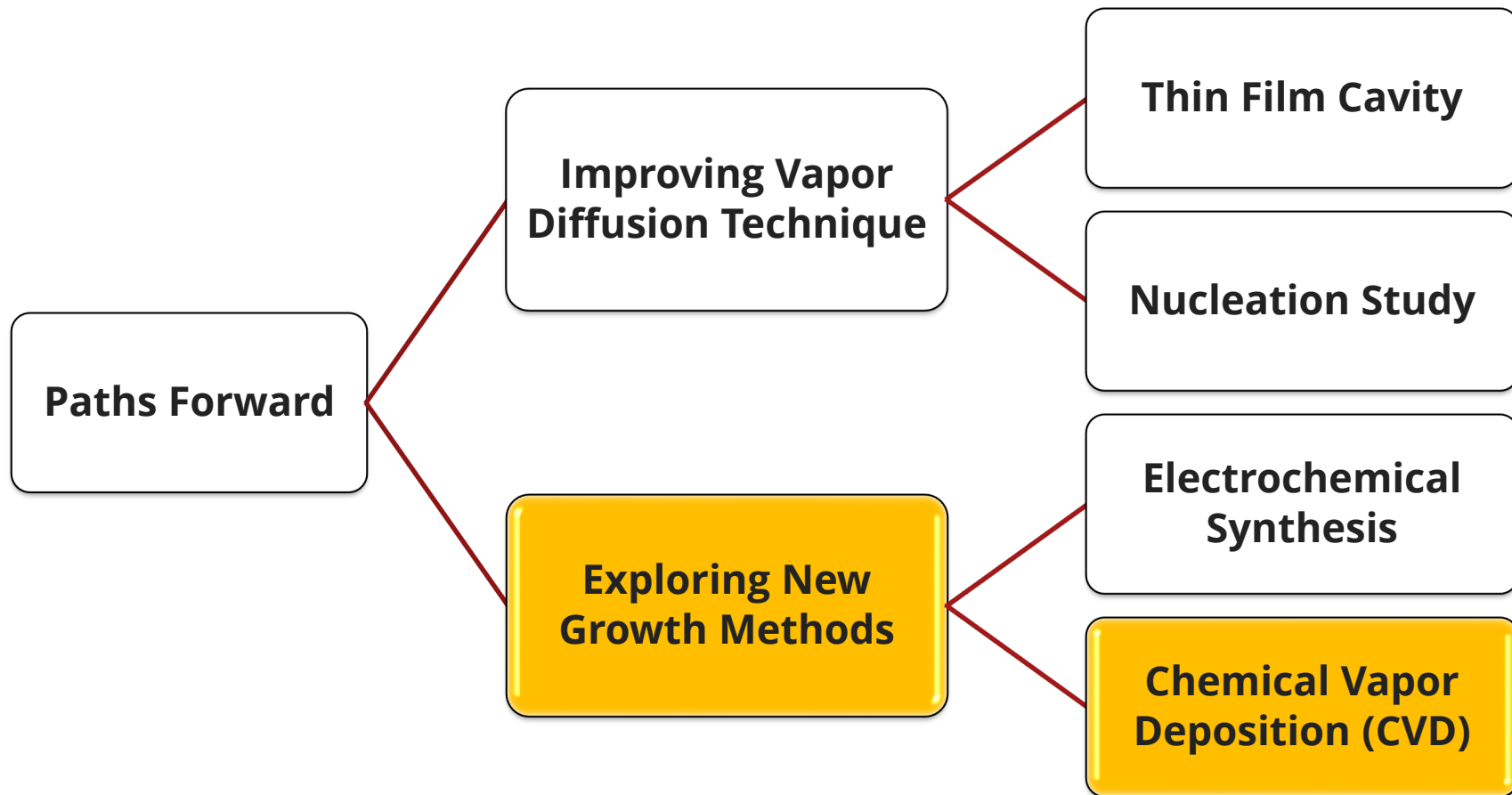
Surface Resistance



Surface Resistance





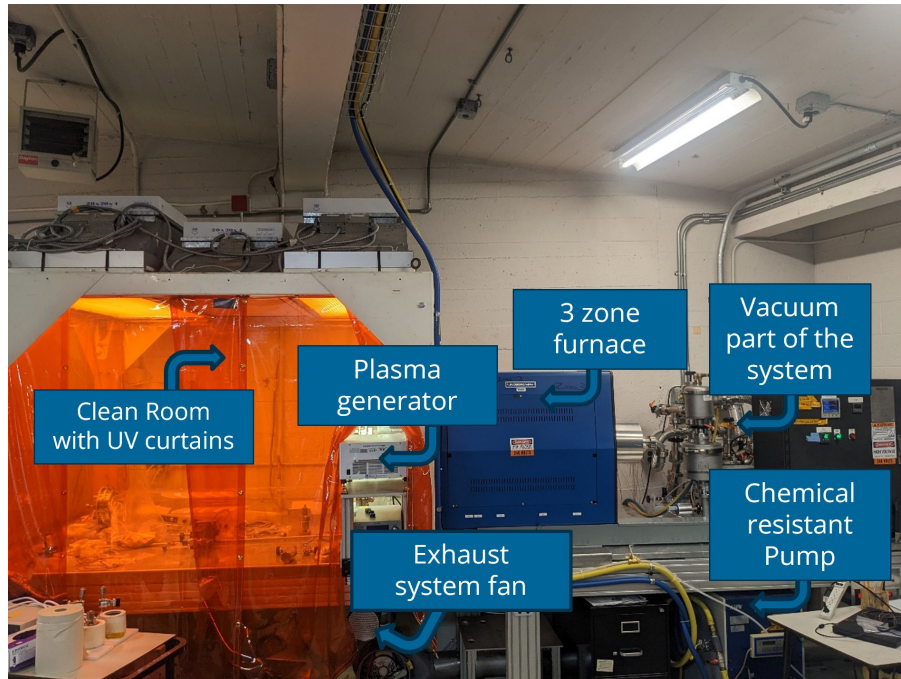
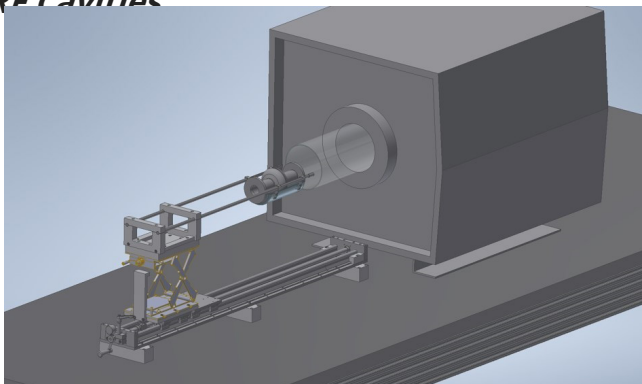


CVD is an alternative method for depositing thin films

→ allows for precise control over the properties of the deposited films

CVD furnace is being commissioned at Cornell University as a potential path to **grow high-quality Nb₃Sn films** (among other SRF materials)

→ See Gabriel Gaitan's poster MOPMB015 *"Development of a Plasma-Enhanced Chemical Vapor Deposition System for High-Performance SRF Cavities"*



Chemical pre-treatments of the niobium oxide **have an effect in the nucleation** of tin in vapor-diffused-based growth \Rightarrow new knob to turn to achieve thinner, smoother films of uniform thickness and composition

Electrochemical deposition (followed by thermal annealing): **First-ever alternative growth method to achieve such high performance** ($Q_0 > 10^{10}$ and low BCS resistance)

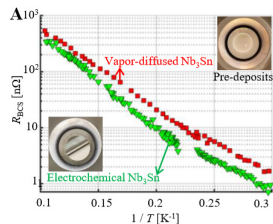
Nb₃Sn Cornell Submissions @ SRF'23

Nucleation Study



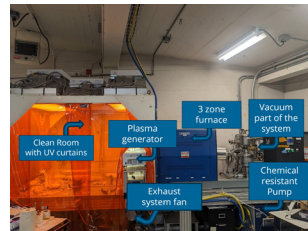
SUSPB026 / MOPMB093

Electroplating



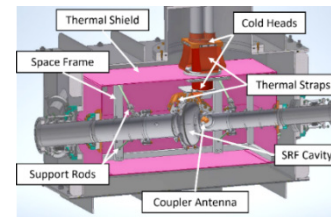
TUPTB006

CVD



SUSPB009 / MOPMB015

Nb₃Sn Cryomodule R&D



SUSPB027, SUSPB028,
THIXA05